

Comparison of Lithium Ion Battery Recycling Processes

2011 Joint US-China

Electric Vehicle and Battery Technology Workshop

August 1-2, 2011

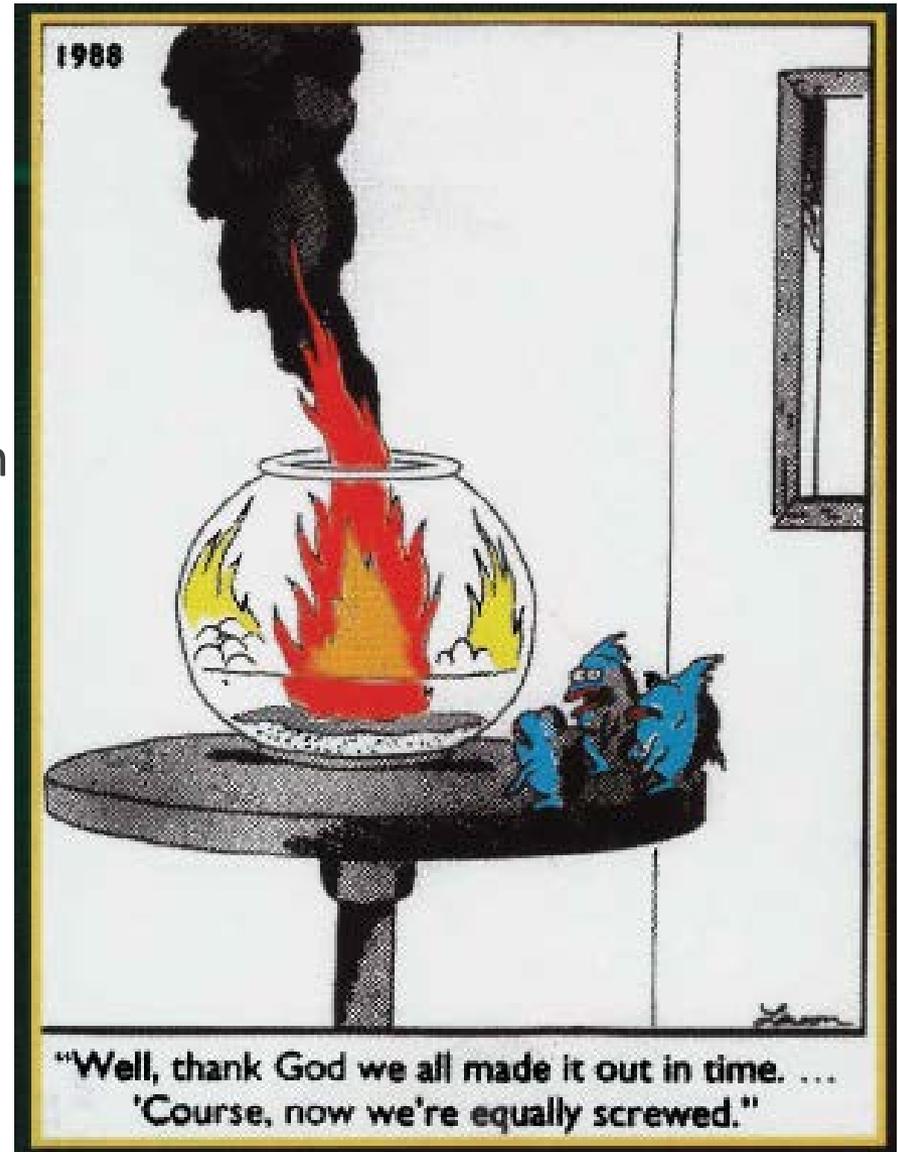
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We don't want to trade one crisis for another!

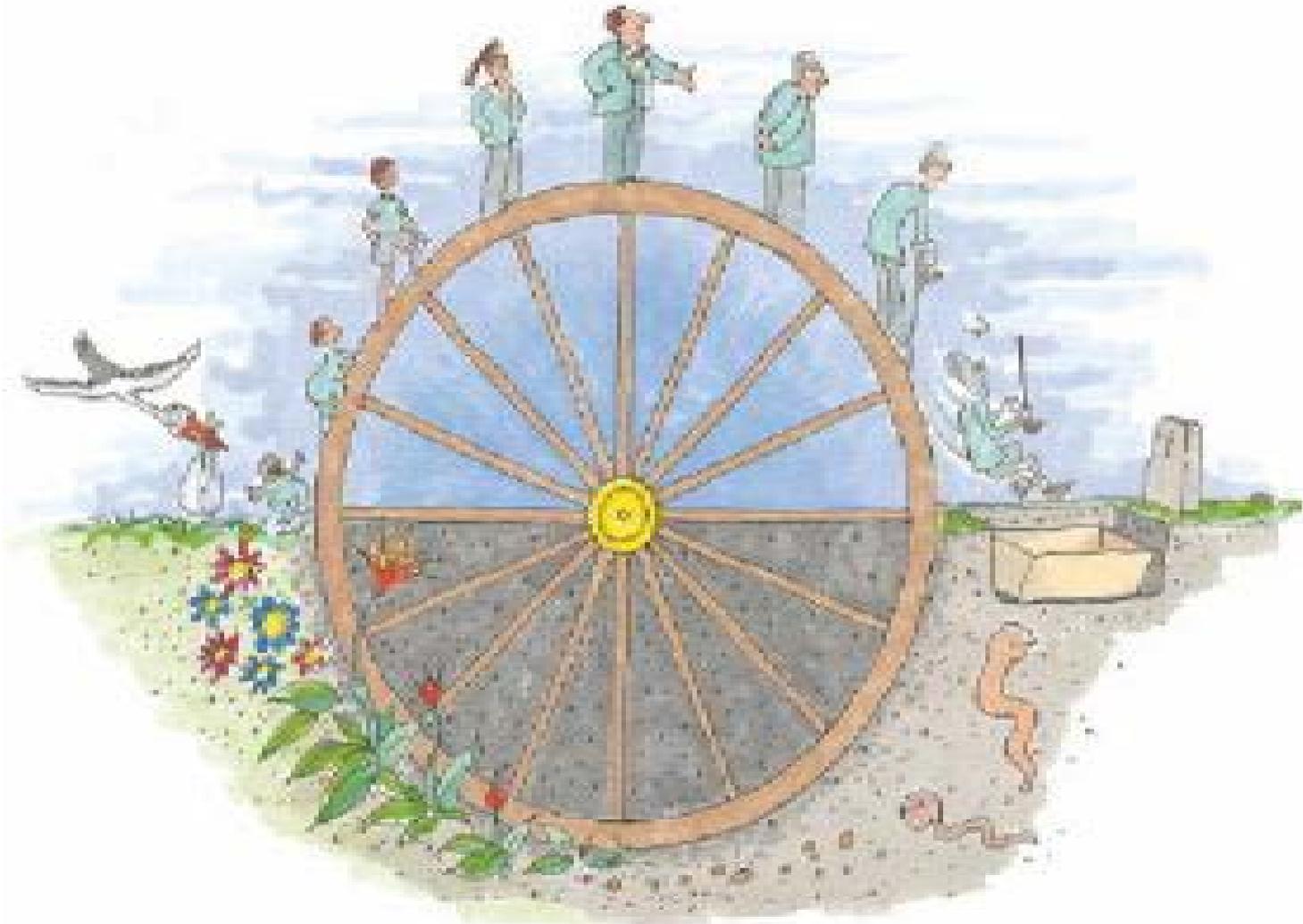
- **Battery material shortages are unlikely**
 - We demonstrated that lithium demand can be met
 - Co supply and price will reduce importance of NCA-G chemistry
 - Scenarios included rapid growth of EVs to 2050
 - Recycling mitigates potential scarcity
- **Now we're checking for unforeseen impacts**

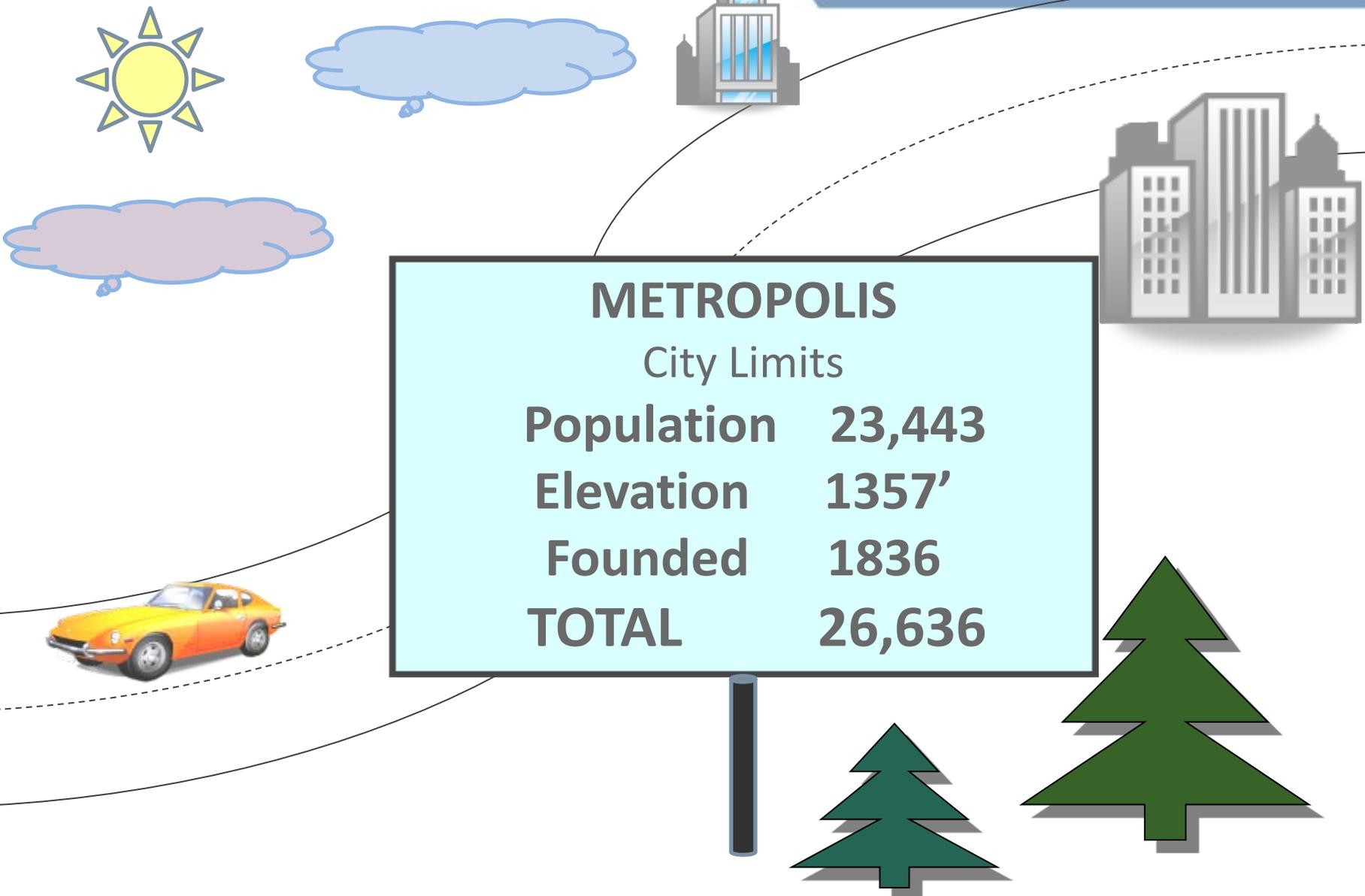


U.S. cobalt use could make dent in reserve base by 2050

Material	Availability (million tons)	Cumulative demand	Percent demanded	Basis
Co	13	1.1	9	World reserve base
Ni	150	6	4	World reserve base
Al	42.7	0.2	0.5	US capacity
Iron/steel	1320	4	0.3	US production
P	50,000	2.3	~0	US phosphate rock production
Mn	5200	6.1	0.12	World reserve base
Ti	5000	7.4	0.15	World reserve base

Lifecycle analysis compares all process impacts of a product's life cycle, from raw material acquisition through production, use, end-of-life treatment, recycling, and final disposal if any.



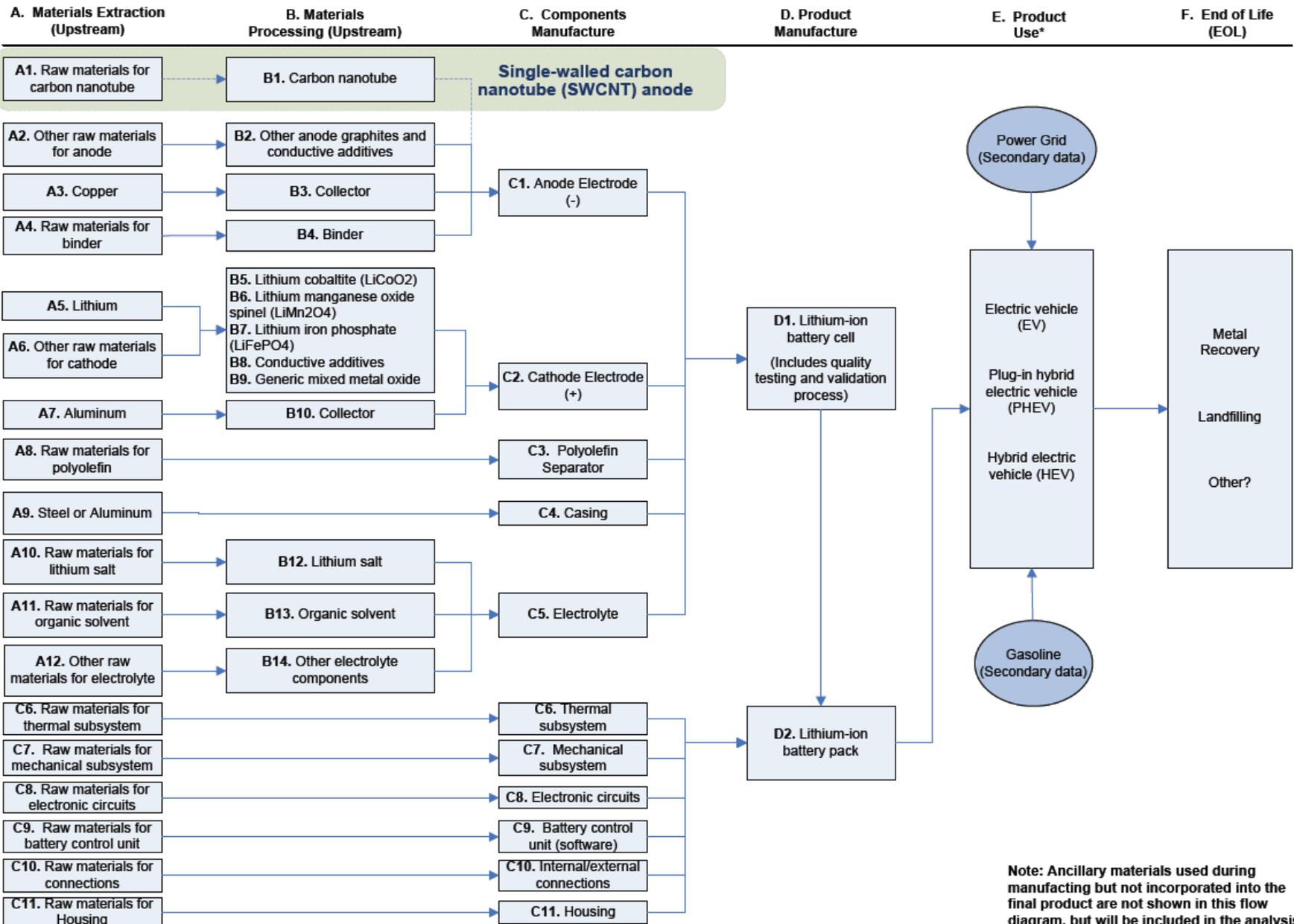


METROPOLIS	
City Limits	
Population	23,443
Elevation	1357'
Founded	1836
TOTAL	26,636

There is no correct way to aggregate impacts into a single "score."



Generic Process Flow Diagram for Passenger Vehicle Energy Systems, Focusing on Lithium-ion Batteries



* The life-cycle assessment will not include impacts associated with manufacturing the vehicles in which the batteries are used.

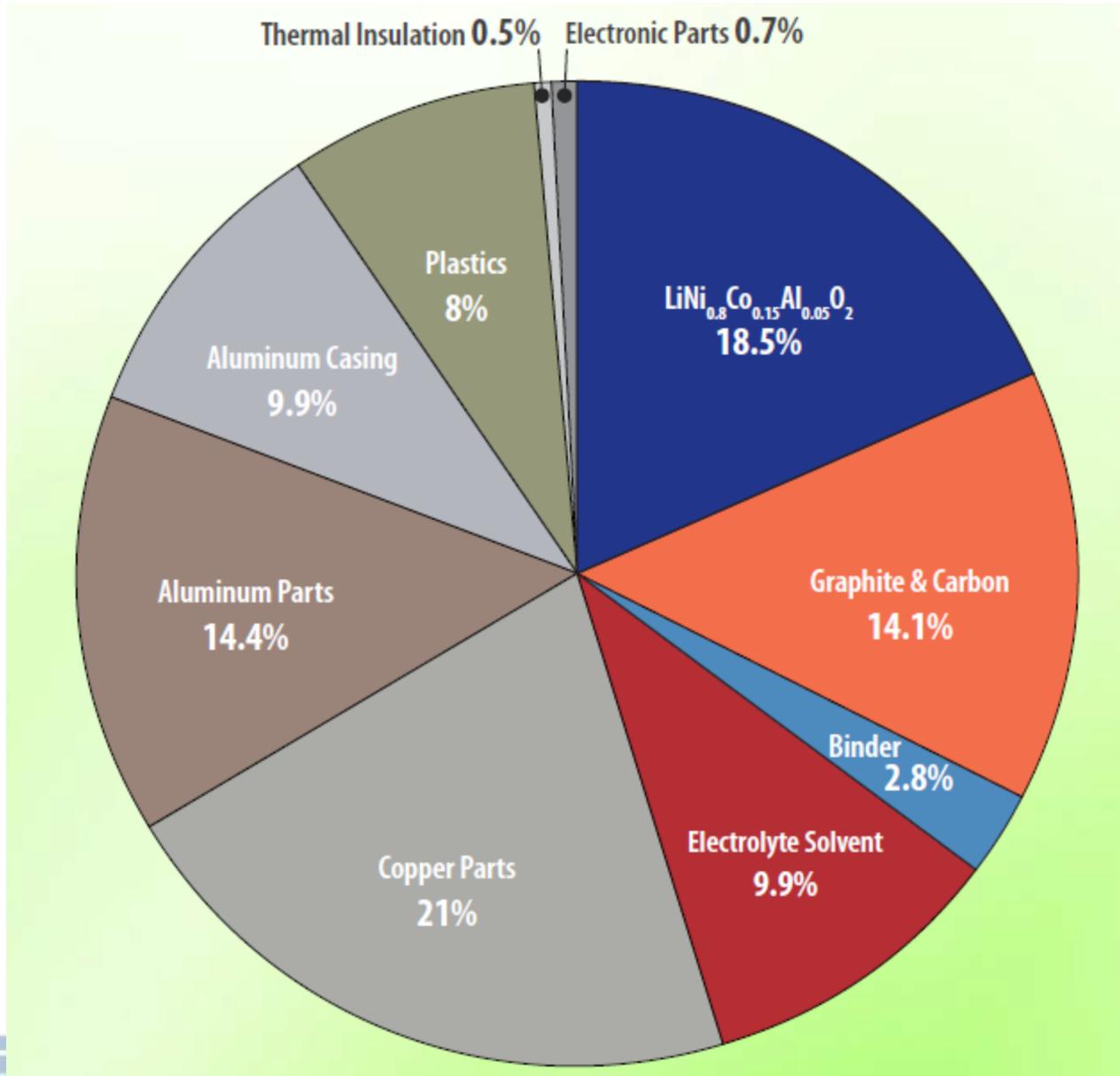
Sources: EPA, Design for Environment, Nanotech Partnership; Components of lithium ion rechargeable batteries, NEC/TOKIN (<http://www.nec-tokin.com>);

Otapiriyakul, S, Caudill, R.J. *A Framework for Risk Management and End-of Life (EOL) Analysis for Nanotechnology Products: A Case Study in Lithium-ion Batteries*. IEEE.

Ganter, M.J, Seagar, TP, Schauerman, CM, Landi BJ, Raffaelle, RP. *A Life Cycle Energy Analysis of Single Wall Carbon Nanotubes Produced Through Laser Vaporization*. Rochester Institute of Technology.

Electropedia, Battery and Energy Technologies, Lithium Ion Battery Manufacturing (http://www.mpoweruk.com/battery_manufacturing.htm accessed on June 7, 2010)

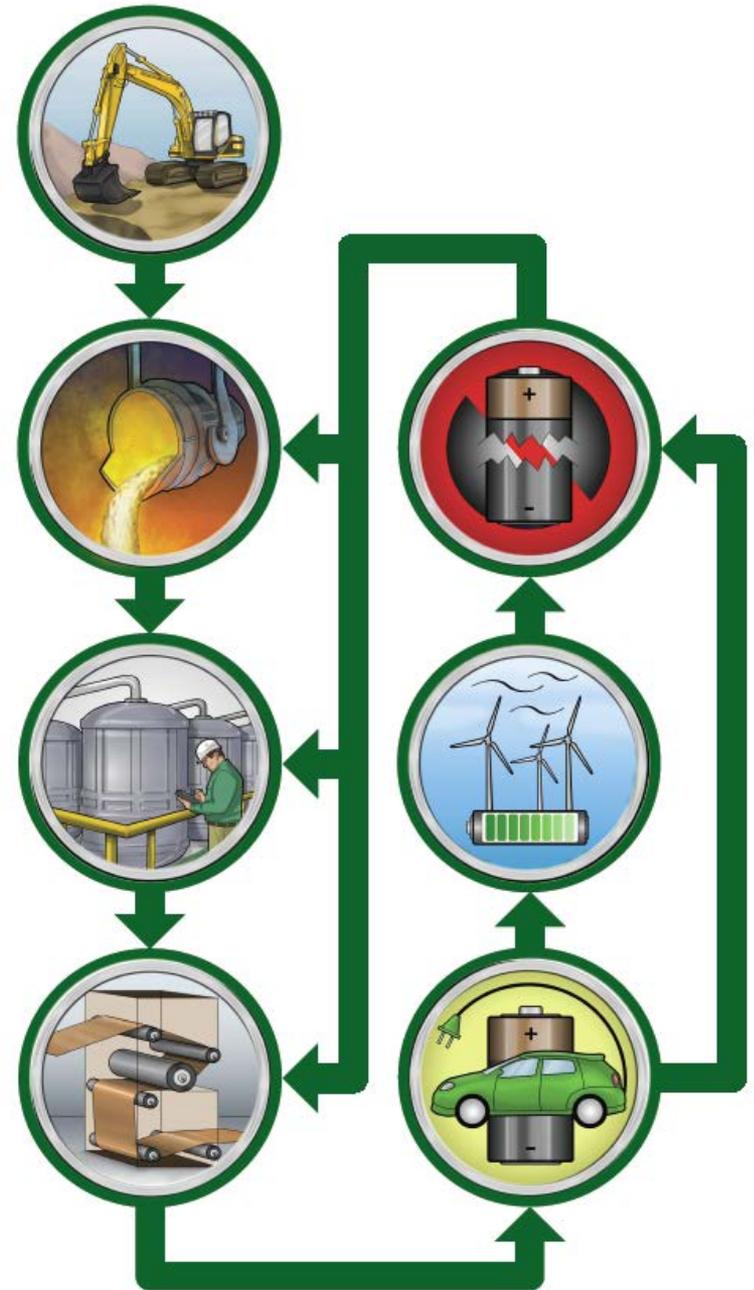
The bulk of battery materials are well characterized; we are now examining the others



Why think about recycling?

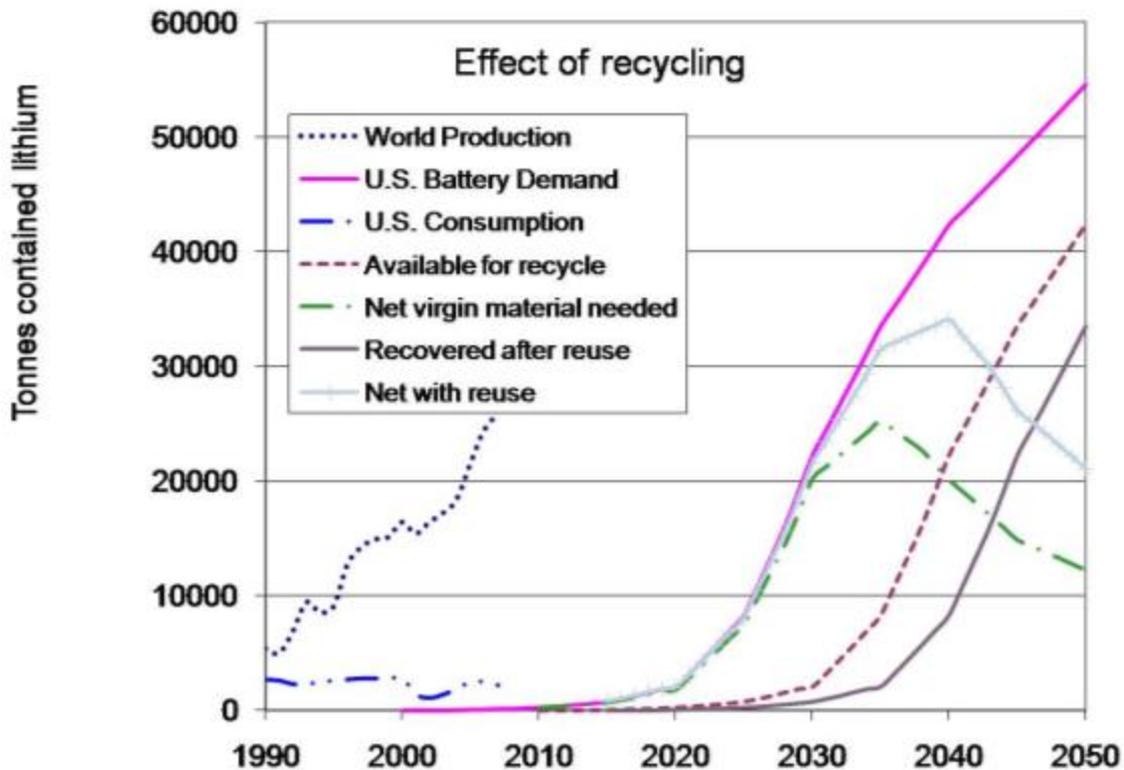
- **Material scarcity alleviated**
- **Recycled materials cheaper**
- **Production impacts avoided**
 - Energy use
 - Emissions
 - Mining impacts
- **Legally required**

- **But not all recycling processes are created equal**
- **Which is best?**
 - LCA identifies “greenest” processes
 - May not be most economical
 - There may be other issues as well



A word about battery reuse

- Reuse takes battery directly back to lower-performance use
- Nissan will reuse batteries to store energy from PV panels or to store back-up power
- Reuse delays return of material for recycling and increases peak demand for virgin material
 - Assumes virgin material would not be used for backup batteries



Recycling reduces production burdens

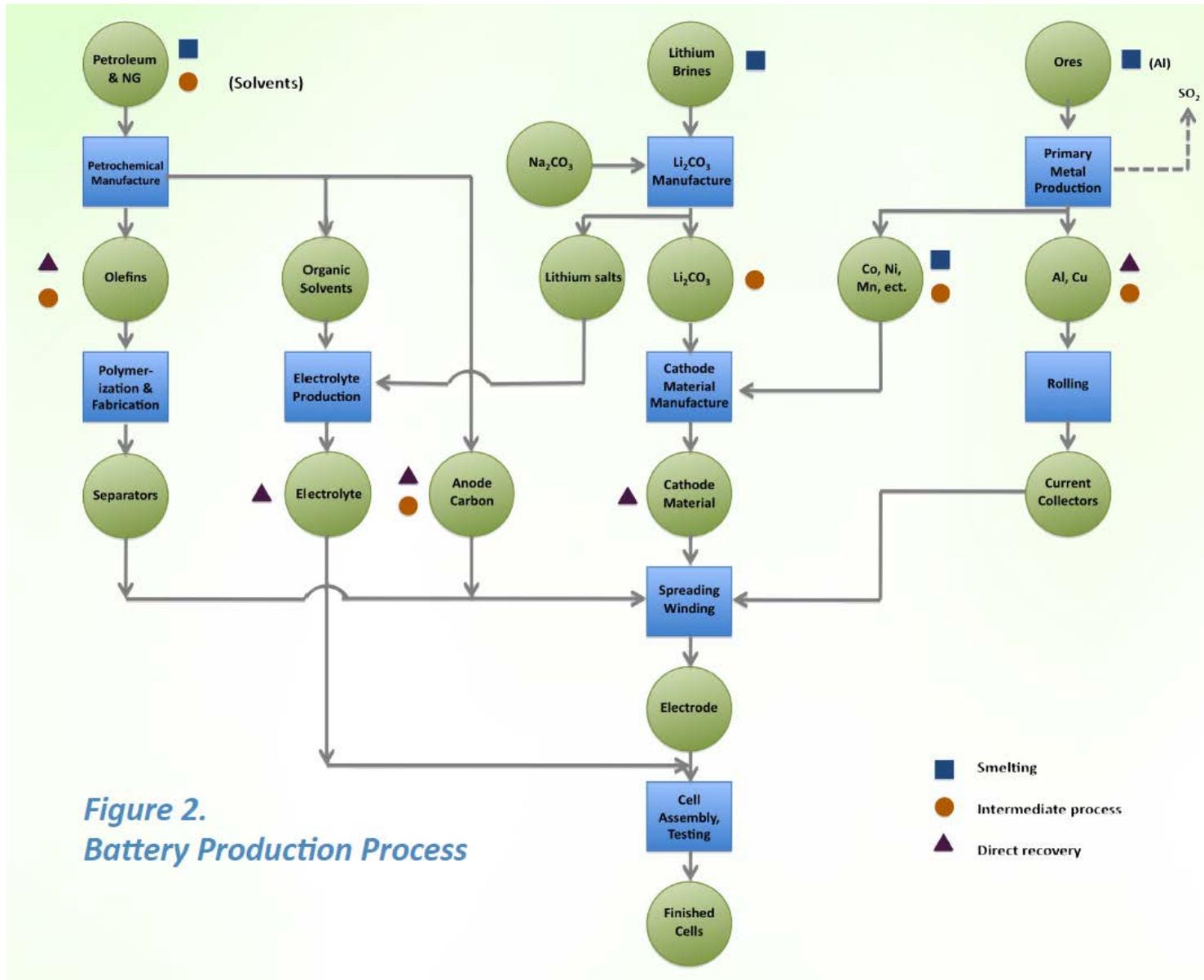
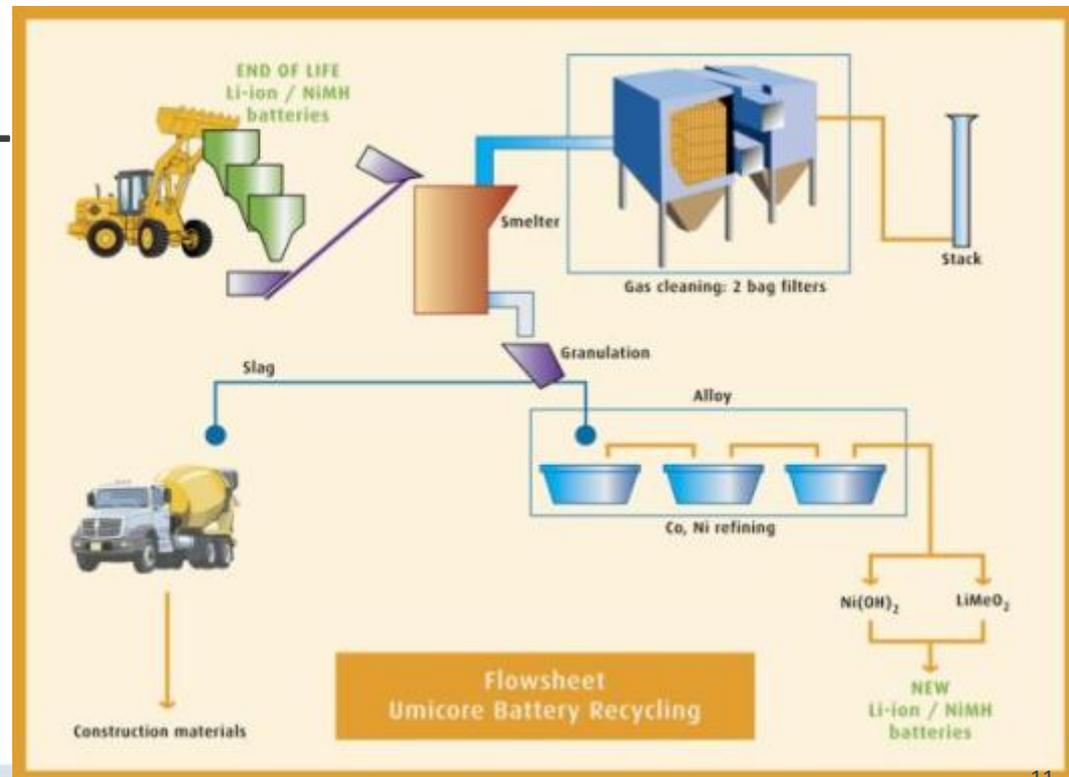


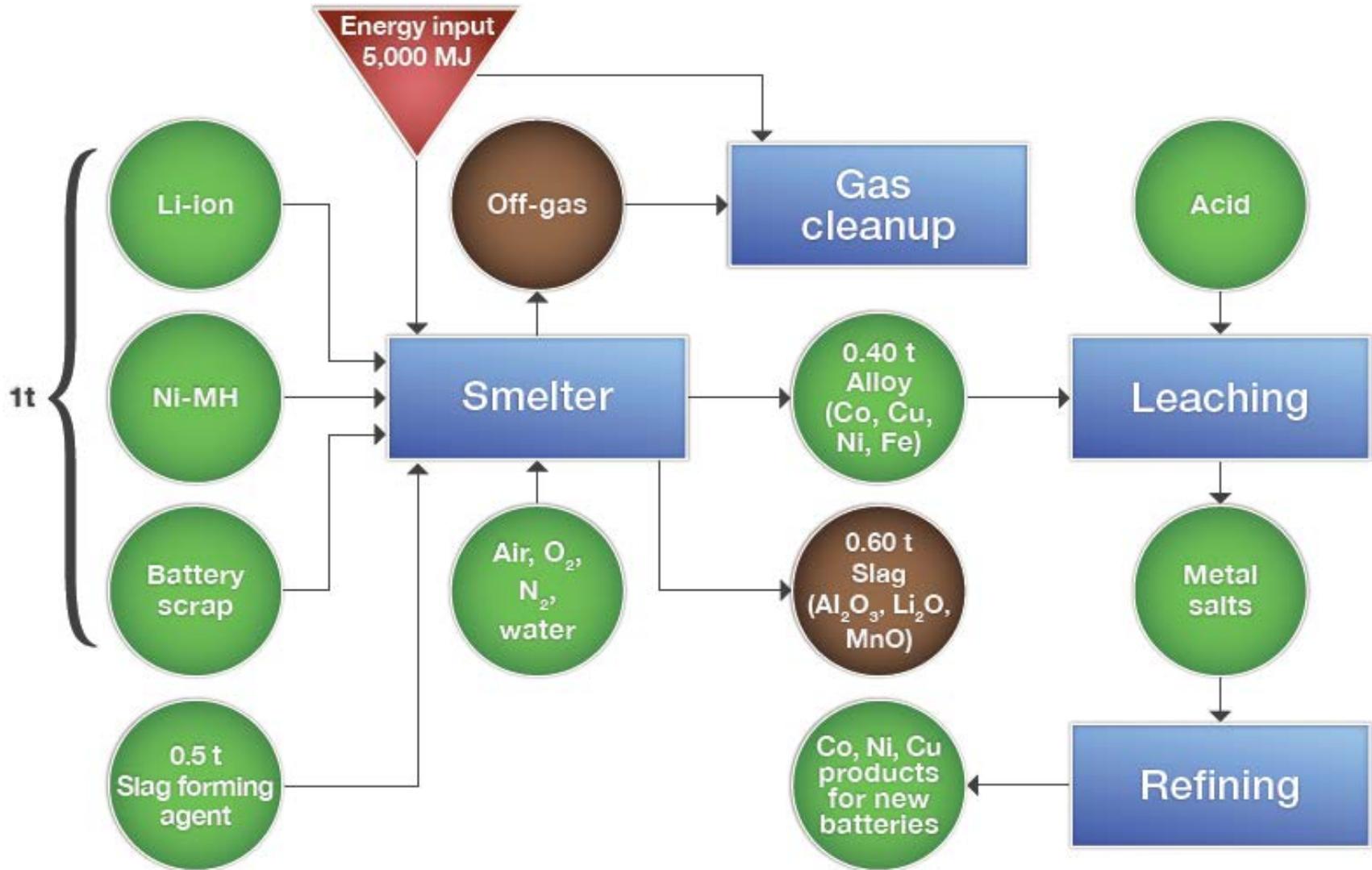
Figure 2.
Battery Production Process

Smelting processes avoid metal ore processing

- These can take just about any input, high volume
- High-temperature required
 - Organics are burned for process energy
- **Valuable metals (Co and Ni) recovered and sent to refining**
 - Suitable for any use
 - 70% of cobalt production energy saved; sulfur emissions avoided
 - Fabrication still needed
 - Less Co → less value
- **Volatiles burned at high-T**
- **Li, Al go to slag**
 - Could be recovered



Commercial smelting process recovers metals



In Finland, batteries are collected and sent to a small plant



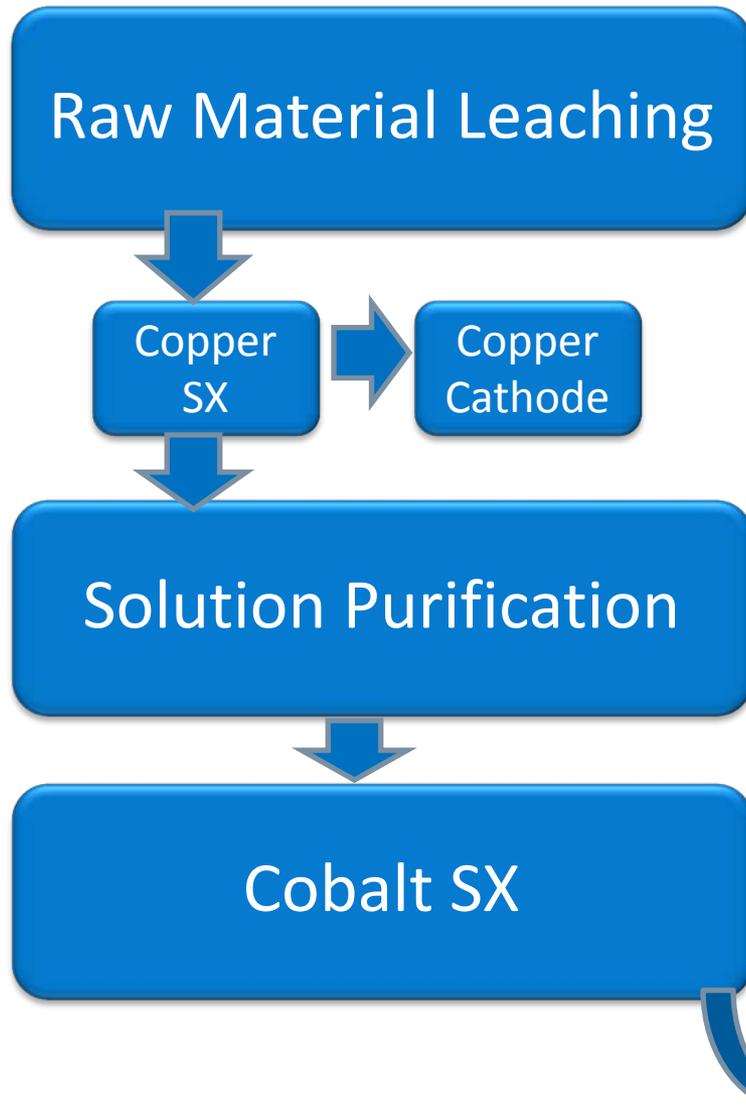
They are sorted and chopped up



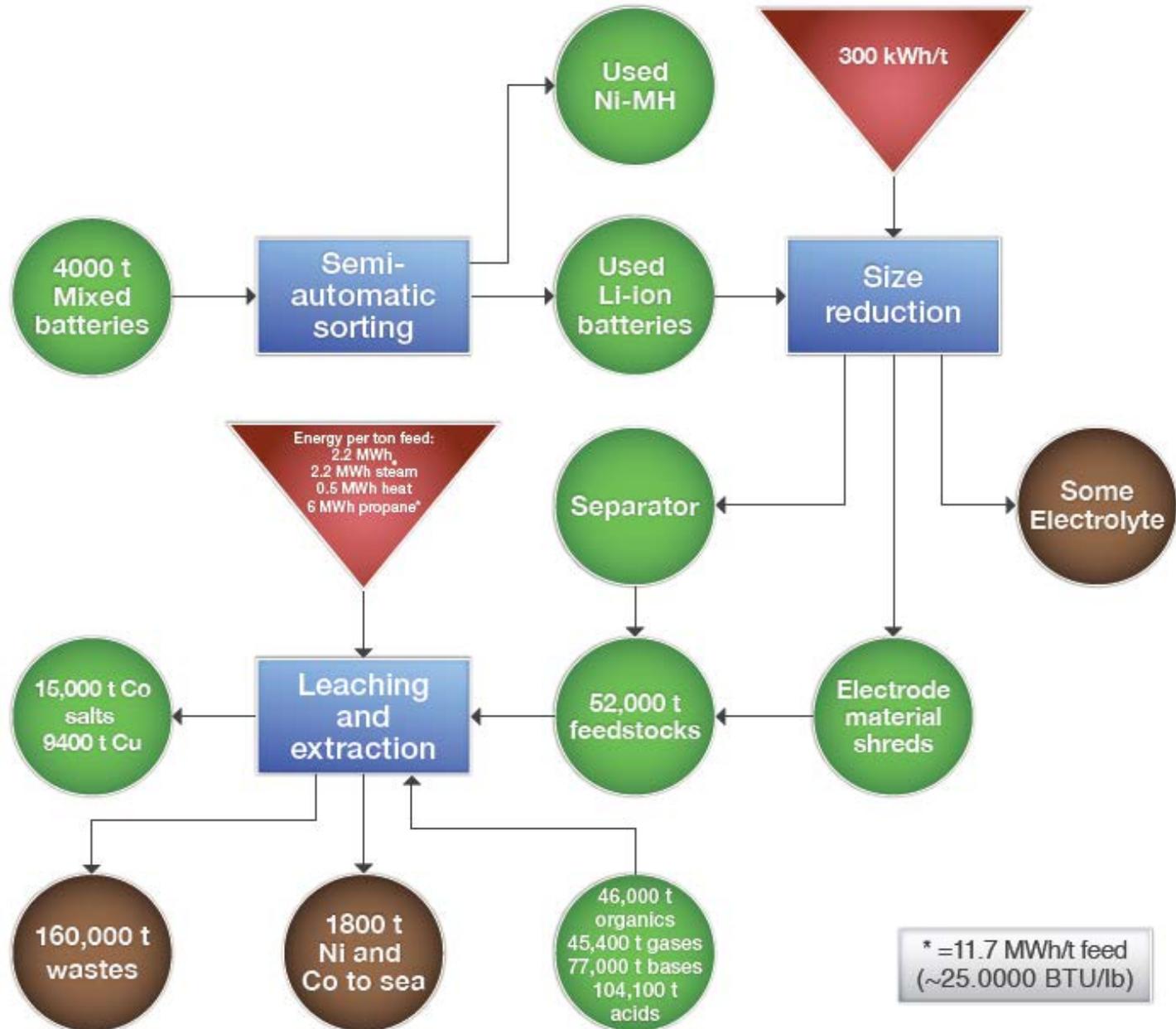
The product is leached at a large plant, along with tailings from a copper mine in Congo to recover cobalt



Several cobalt products are produced



Battery materials are small part of cobalt plant input



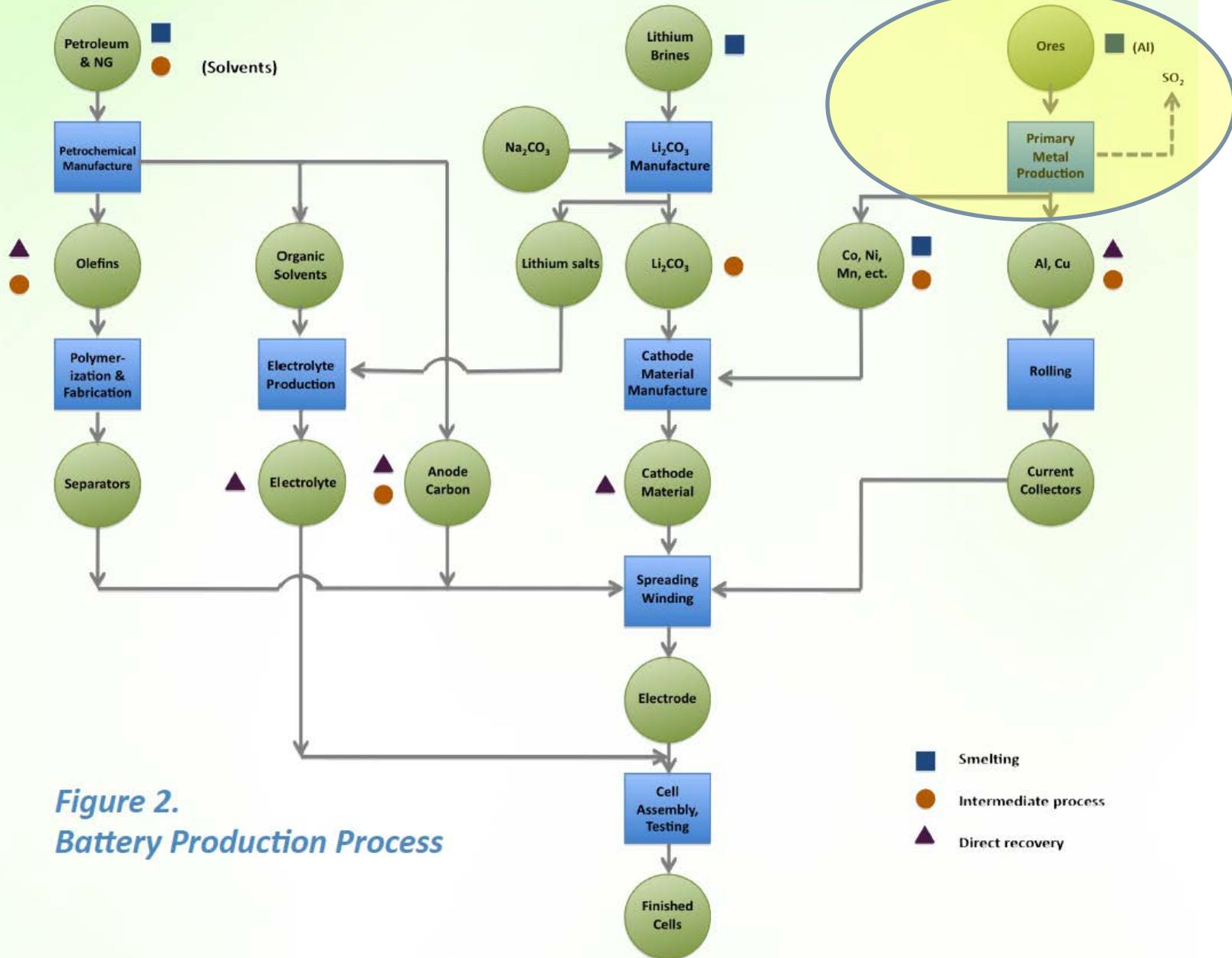
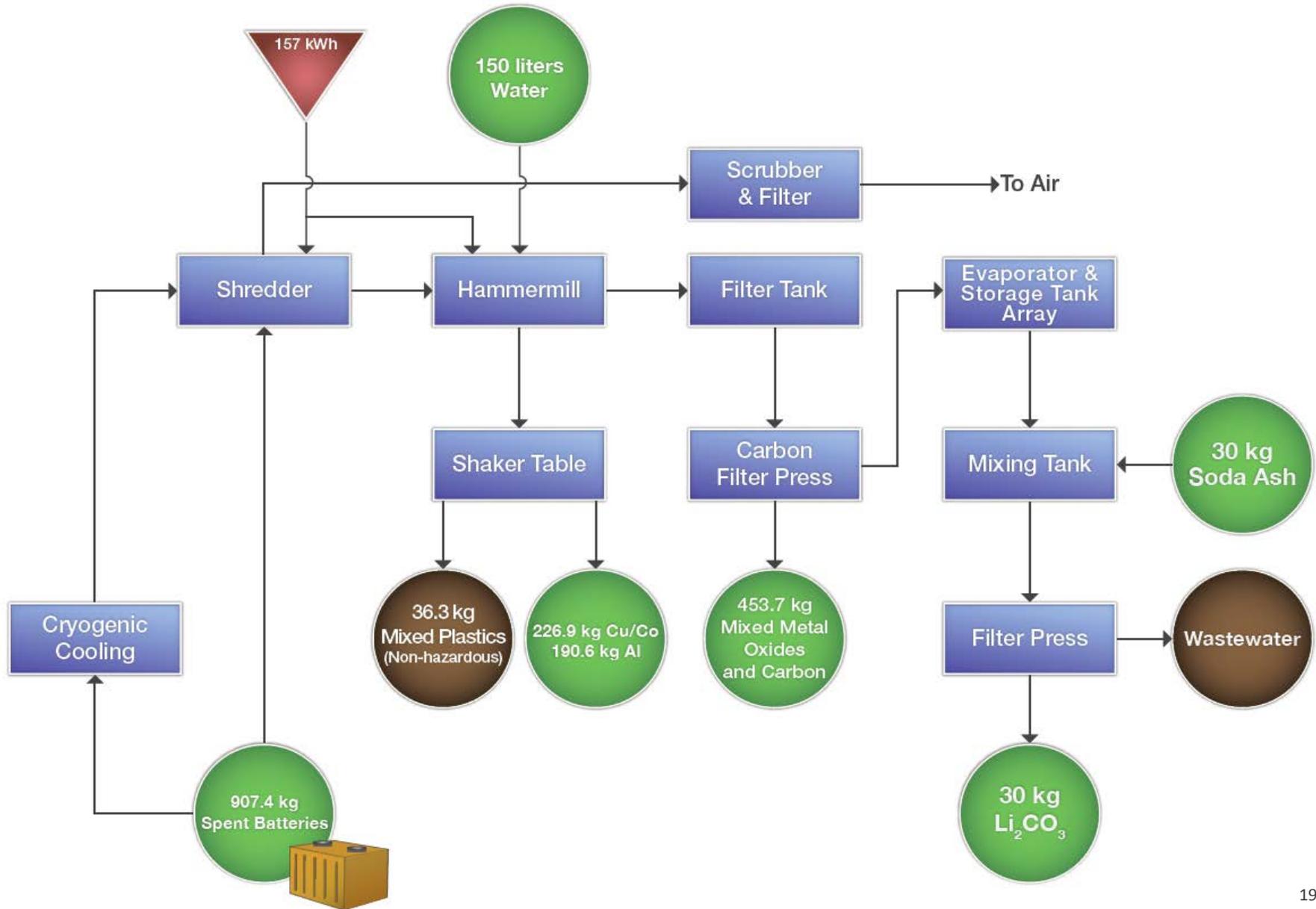
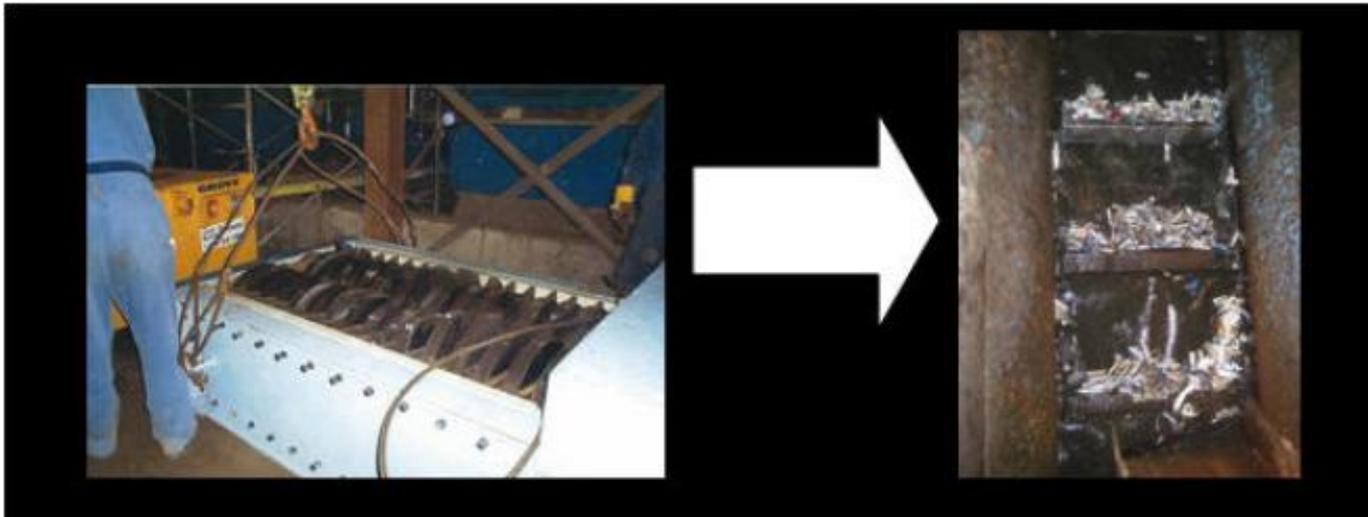


Figure 2.
Battery Production Process

Toxco Ohio plant will use improved process to recover lithium as well as metals



Products are suitable for further processing



Copper cobalt and filter cake products

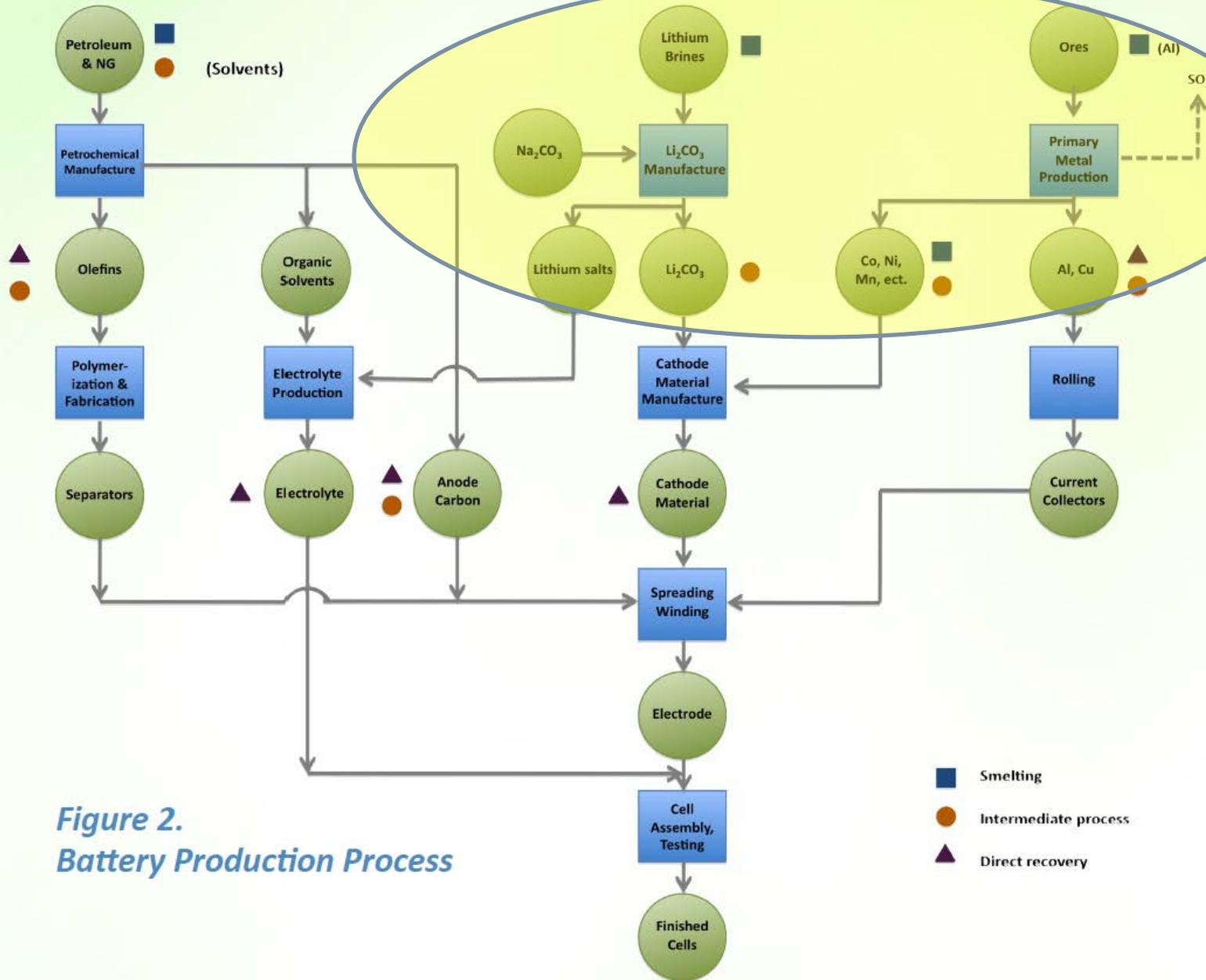
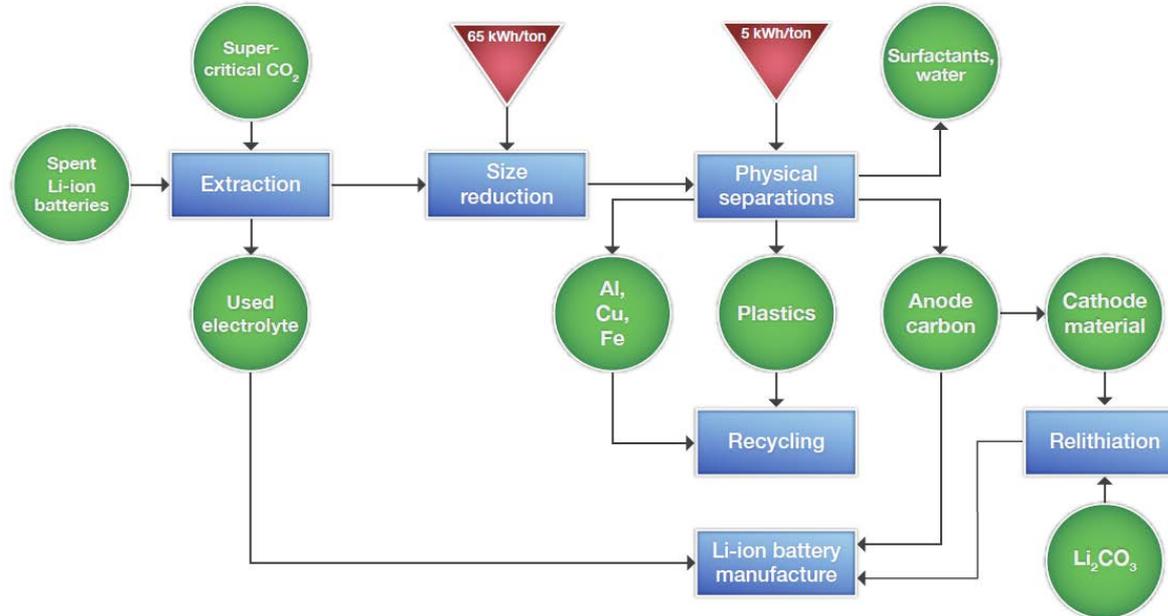


Figure 2.
Battery Production Process

Recovery of battery-grade materials avoids impacts of production from raw materials

- Requires as uniform feed as possible
- Components are separated to retain valuable material structure
 - Costs lower than virgin materials
 - Purify/reactivate components if necessary for new batteries
 - Separator is unlikely to be usable, as form cannot be retained
- Low-temperature process, low energy requirement
- Does not require large volume



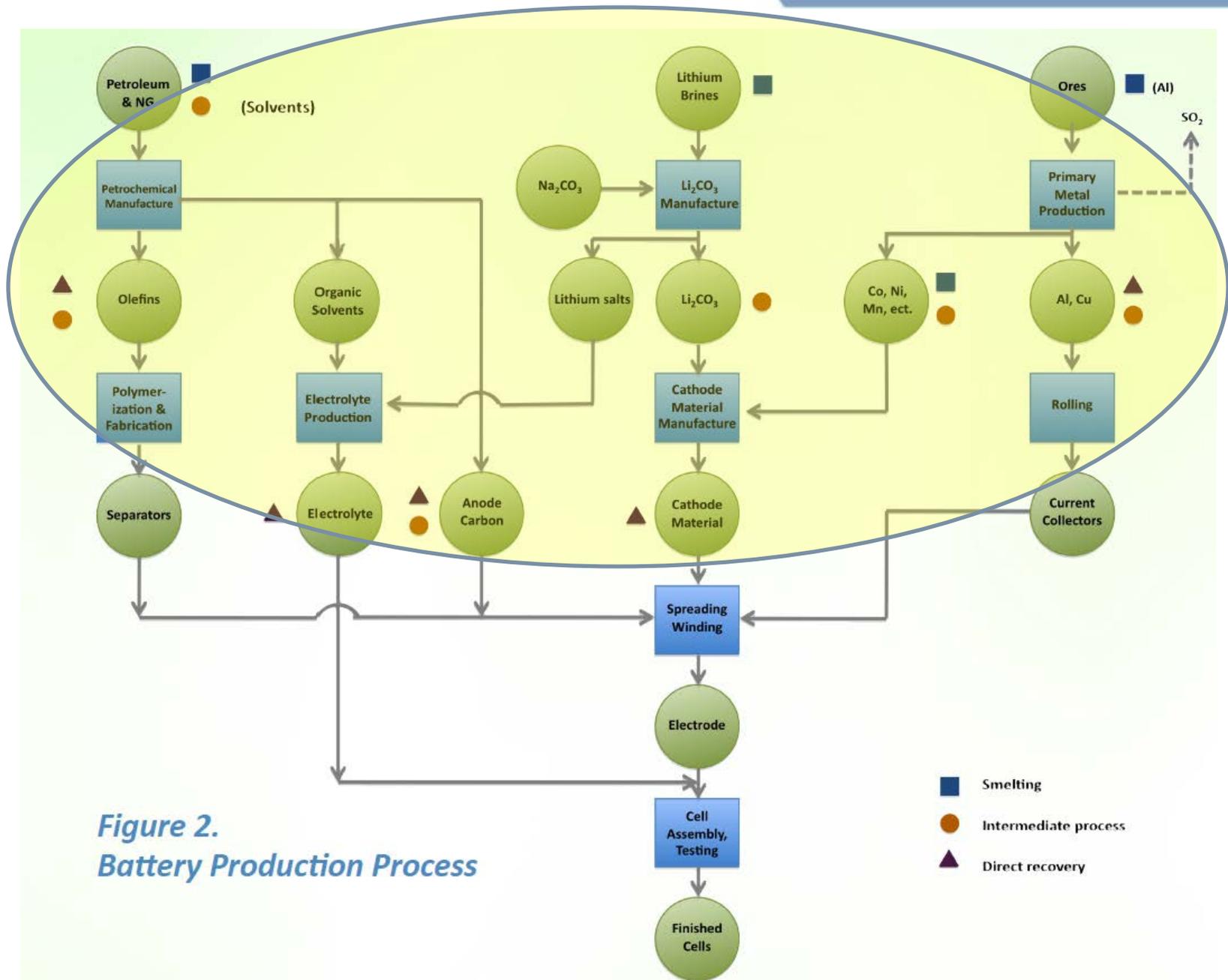


Figure 2.
Battery Production Process

Plastics can be made into C-nanotubes

- Argonne has developed a process for plastic bags
 - Vilas Pol is developer
 - Process could be used for battery plastics
- React with cobalt acetate catalyst at 700° C, cool 3 hours
 - Recover catalyst when battery recycled
- C-nanotube anodes are produced
 - Now made from petroleum at ~\$100/gram
 - Recycling process would be cheaper
- Argonne solvent extraction process also could be used to recover battery plastics
 - Could utilize plastic stream from Toxco/Kinsbursky or Eco-Bat



Battery materials are responsible for only 2% of energy but 20% of life-cycle SOx!

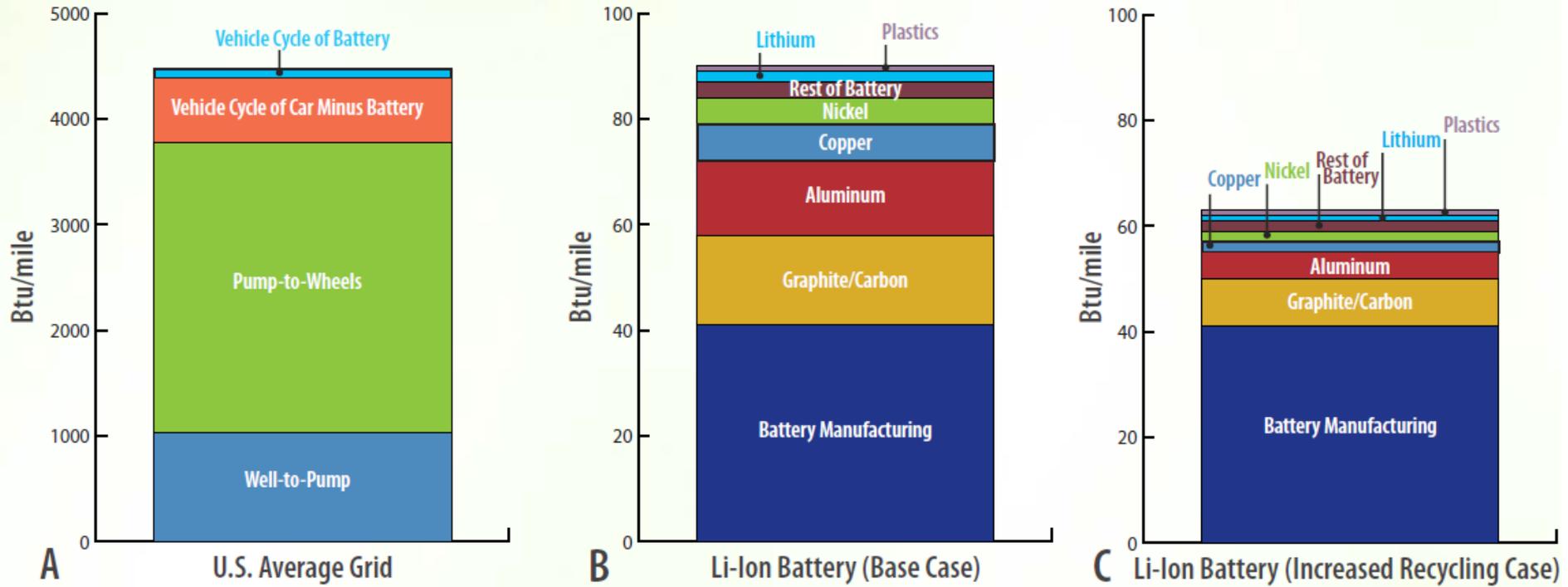
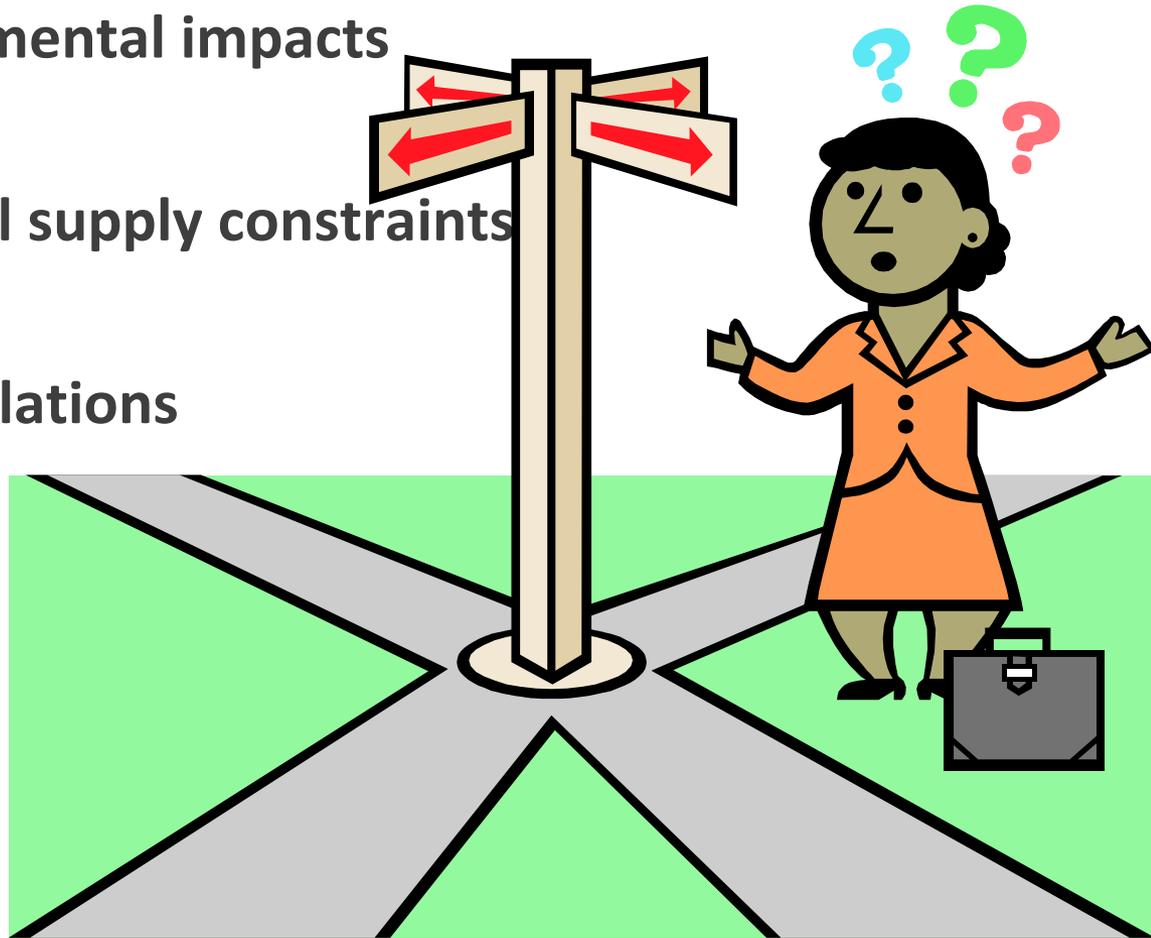


Figure 3. Life-Cycle Energy Analysis of PHEV20

Battery impacts rise with all-electric range.

Recycling processes will be compared using several interconnected criteria

- Reduction of environmental impacts
- Energy savings
- Alleviation of material supply constraints
- Economics
- Compliance with regulations
- Feed requirements
- Utility of products
- Scale



Several strategies could facilitate recycling

- **Standard configuration enables design of recycling equipment**
- **Chemistry standardization reduces need for battery sorting**
- **Cell labeling enables sorting**
- **Design for disassembly enables material separation**



Stay tuned for more results. Thank you!

- Co-authors: P. Nelson, J. Sullivan, A. Burnham, and I. Belharouak
- Work sponsored by DOE Office of Vehicle Technologies
- Contact me: lgaines@anl.gov

