

Species Time-Histories and Temperature Sensing at Elevated Pressures

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- **Shock Tube Kinetics**
 - **Experimental Approach**
 - **Ignition Delay Times**
 - **Species Time Histories**
- **Temperature Sensing**
 - **Experimental Approach**
 - **CO₂-Based T Sensor**
 - **H₂O-Based T Sensor**
 - **Application Domains**

Some of the work presented here is unpublished. Please check with RKH/DFD before regarding the data as final or importing it into publications. We would also value feedback from team members regarding our data and how it might be modeled.

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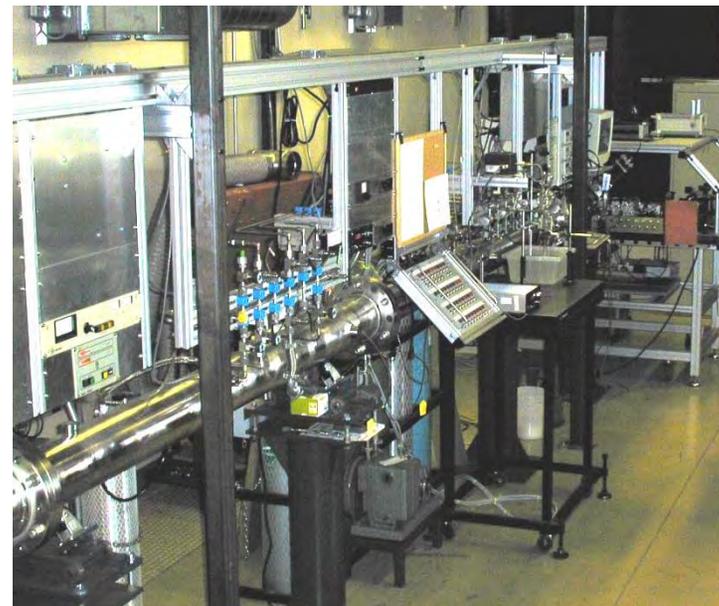
Shock Tube Kinetics: Experiment Types

Stanford STs provide access to pressures from 1 to 500+ atm

- HP Ignition delay times provide global targets for engine models
- **Species time-histories provide strong constraints on mechanisms**
 - **CW Laser absorption can provide species time-histories for: OH, CO, CO₂, CH₂O, H₂O, CH₃, CH₄, C₂H₄, MF, MeOH, NH₃, fuel, ...**
- Direct determination of elementary reaction rate constants
 - Provide k where estimates/theory are not sufficient
 - Provide access to high-pressure limit: k_{∞}

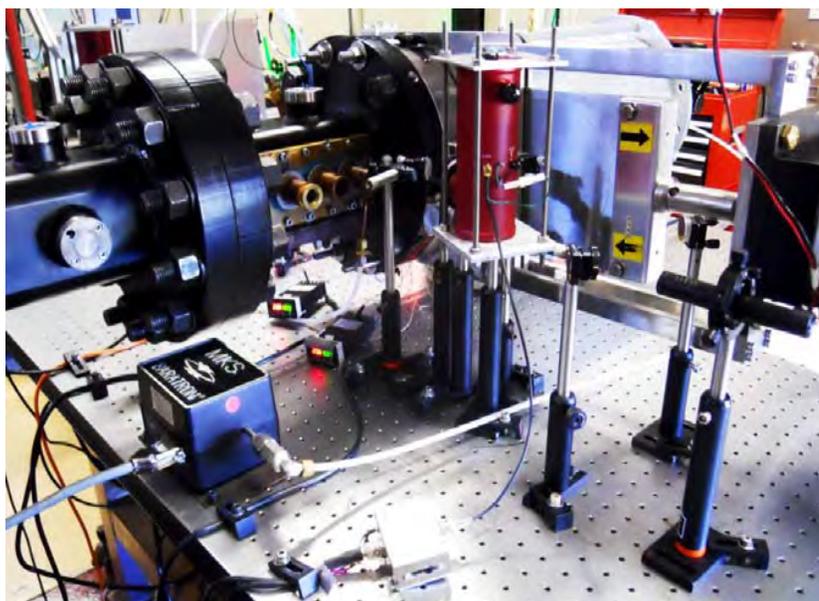


**Kinetics
Shock
Tube 1
(30 atm)**

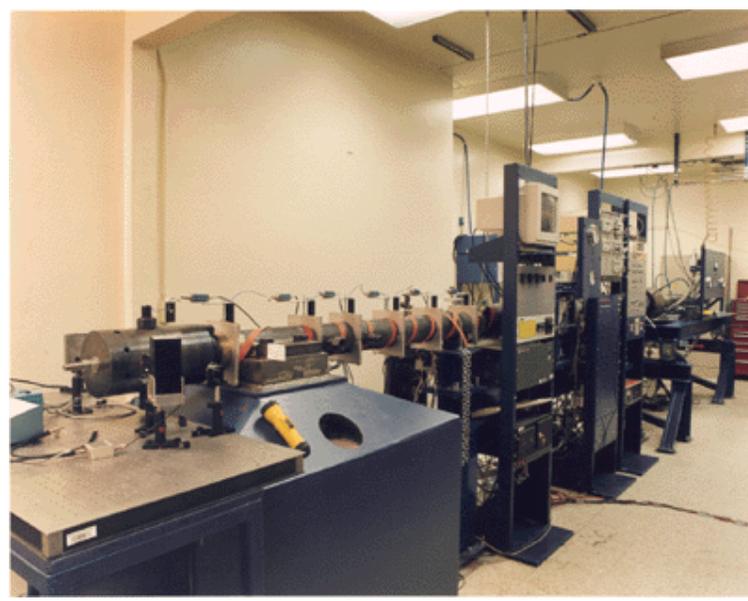


**Kinetics
Shock
Tube 2
(30 atm)**

Experimental Approach



**Aerosol
Shock
Tube
(10 atm)**



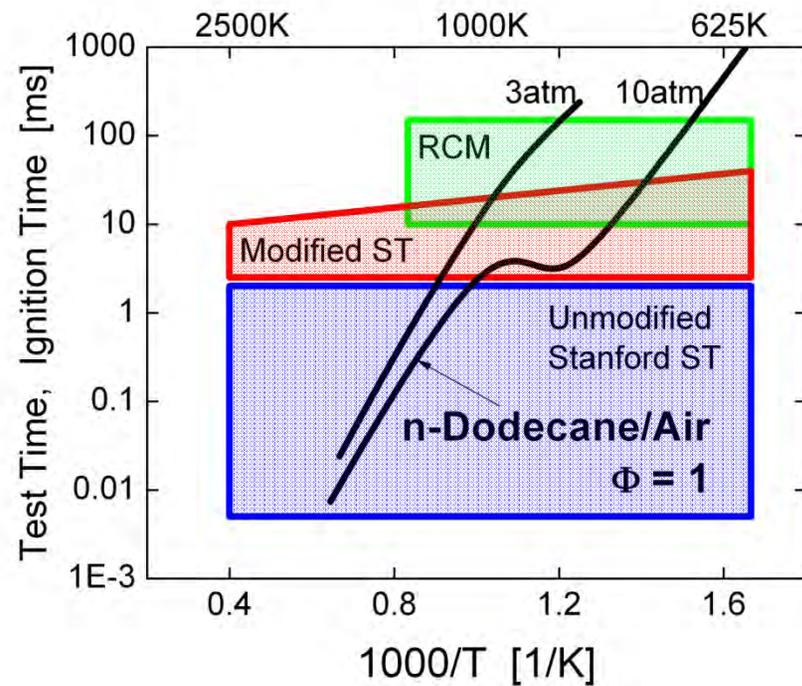
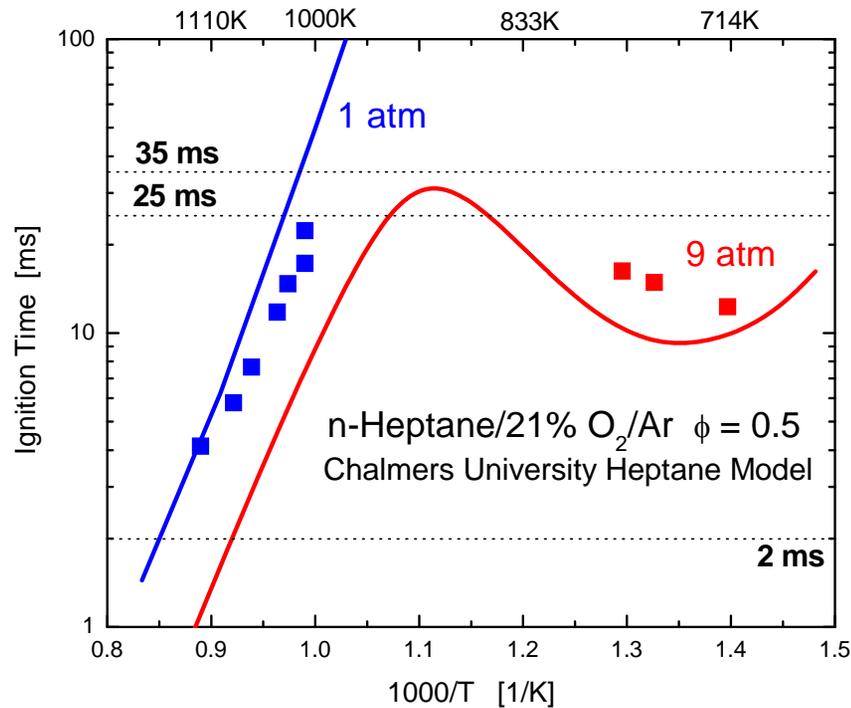
**High
Pressure
Shock
Tube
(500 atm)**

Advances in Shock Tube Methodology

- **Tailored driver gas/new driver geometry** provide extended test time for access to high P and low T
- **Driver inserts** provide highly uniform reflected shock conditions approaching constant U/V
- **Aerosol shock tube** provides access to low-vapor-pressure fuels
- **Gasdynamic modeling of shock tube flows** to account for facility effects and energy release

Access to High Pressure & Low Temperatures

- Longer driver length and tailored gas mixtures can provide longer test times (> 40 ms)

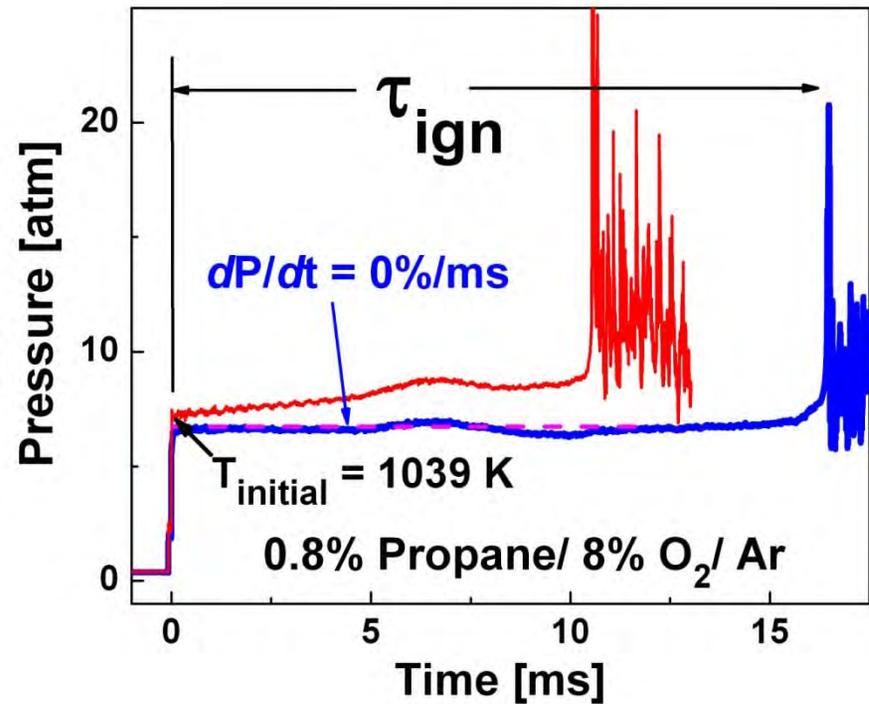


- Shock tubes now can overlap with RCMs

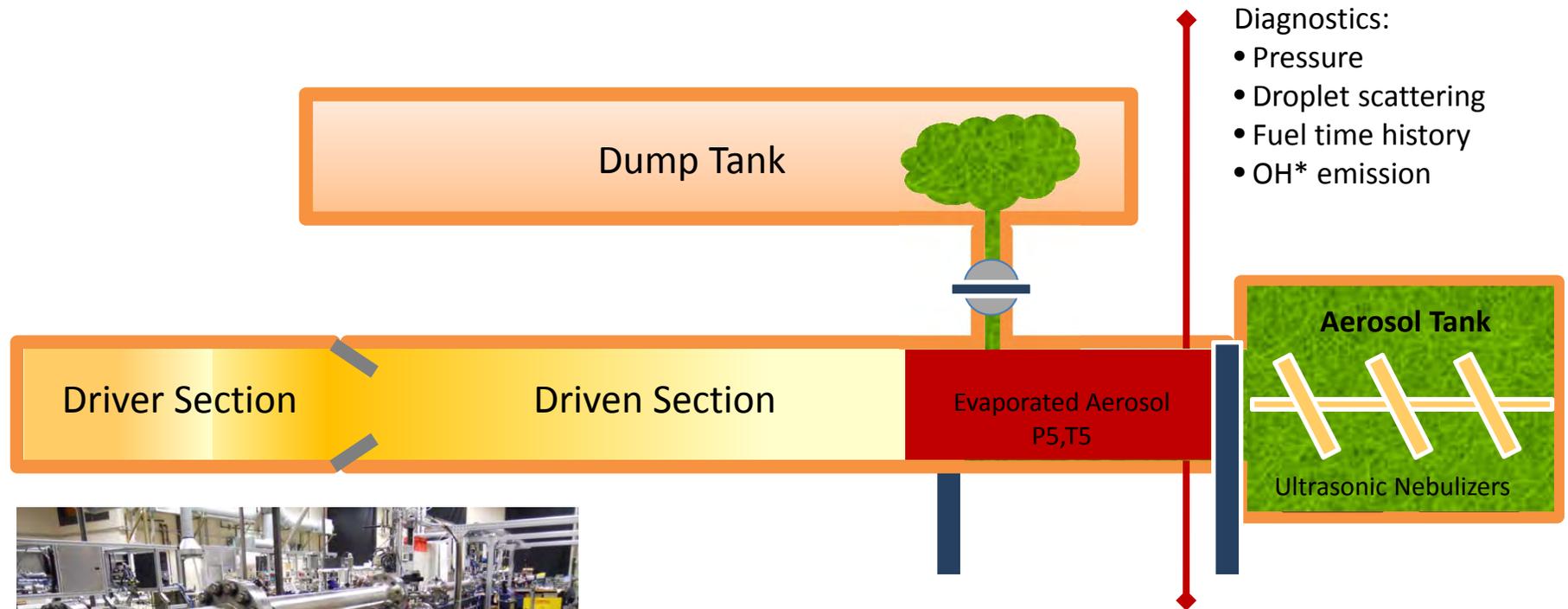
Improvement in Pressure & Temperature Uniformity

- **Problem:** Ignition delay times are artificially shortened by non-ideal facility effects!! $dP/dt \neq 0$
- **Solution:** Driver Inserts
- **Results:** Near-ideal constant-volume performance!! $dP/dt \approx 0$

$dP/dt \approx 0$ for $t > 17$ ms



Aerosol Shock Tube for Low-Vapor-Pressure Fuels



- Does not require heated shock tube
- Eliminates fuel cracking and partial distillation
- Provides access to low-vapor-pressure fuels:
large methyl esters, bio-diesel surrogates

Absorption Spectroscopy at High Pressures

Beer-Lambert Law (for absorption of monochromatic light)

- $T_\lambda = I/I_0 = \exp(-N_i \sigma L) = \exp(-\{X_i P/RT\} \sigma L)$
- **A** = absorption = $(1 - I/I_0) \sim \{X_i P/RT\} \sigma L$

At High Pressures

a) center of isolated lines

Hence **A** $\sim X_i L$

$$\sigma_{\text{peak}} \sim 1/\Delta\nu \sim T/P$$

independent of P!

b) broad spectral features

Hence **A** $\sim X_i P L/T$

$\sigma \approx \text{constant}$

increases with P!

Current Laser Capabilities for Species Detection for real-time, *in situ* sensing

Ultraviolet

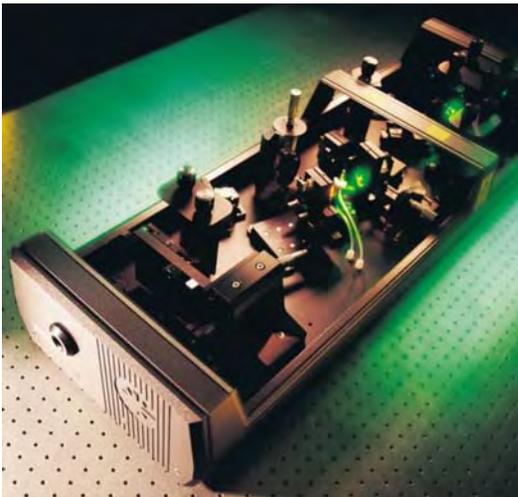
CH ₃	216 nm
NO	225 nm
O ₂	227 nm
HO ₂	230 nm
CH ₂ O	305 nm
OH	306 nm
NH	336 nm

Visible

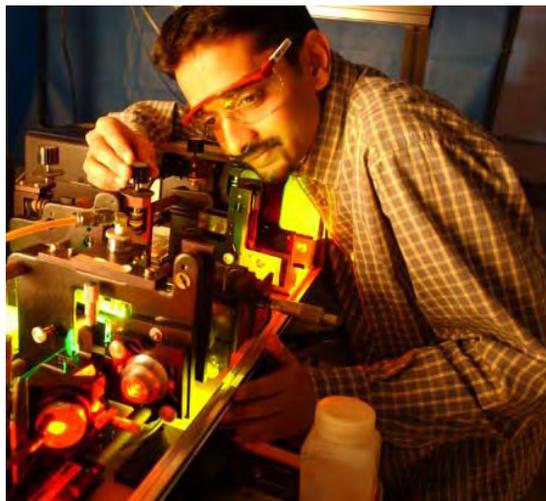
CN	388 nm
CH	431 nm
NCO	440 nm
NO ₂	472 nm
NH ₂	597 nm
HCO	614 nm

Infrared

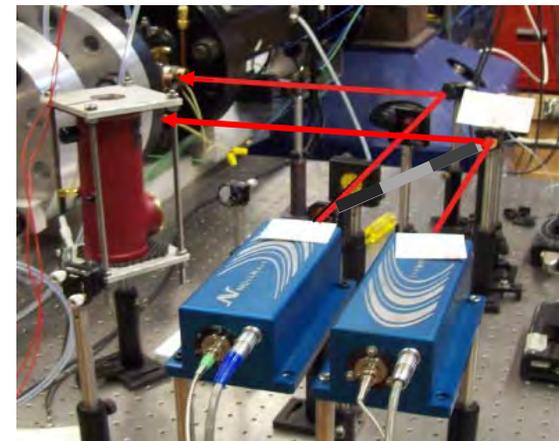
H ₂ O	2.5 μm
CO ₂	2.7 μm
Fuel	3.4 μm
CO	4.6 μm
NO	5.2 μm
MF, MeOH	9.2, 9.7 μm
NH ₃ , MMH	9.6, 10.2 μm
C ₂ H ₄	10.5 μm



Coherent MIRA Ti-Sapphire



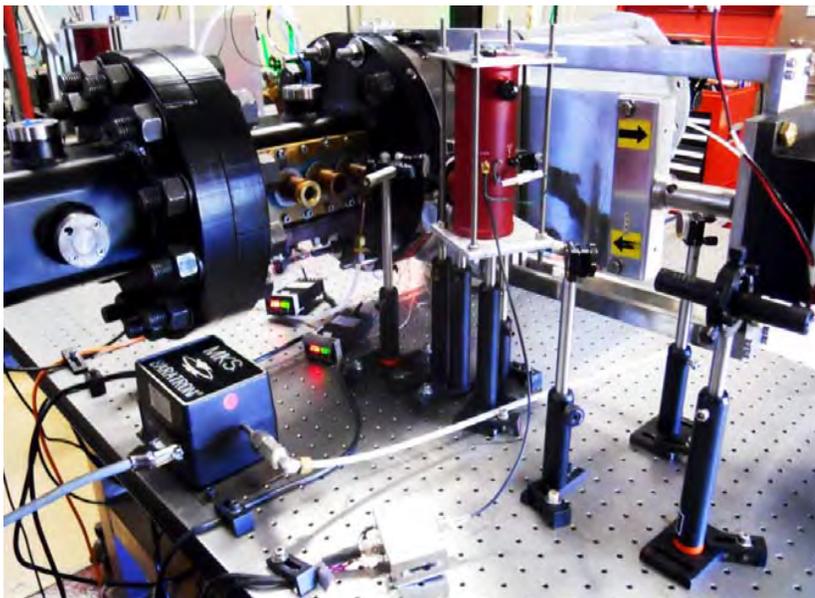
Spectra-Physics 380 Ring



NovaWave Mid-IR DFG



HP Ignition Delay Time Data



**Aerosol
Shock Tube
10 atm**

**High
Pressure
Shock Tube
500 atm**



HP Ignition Delay Time Data

2 Representative Examples

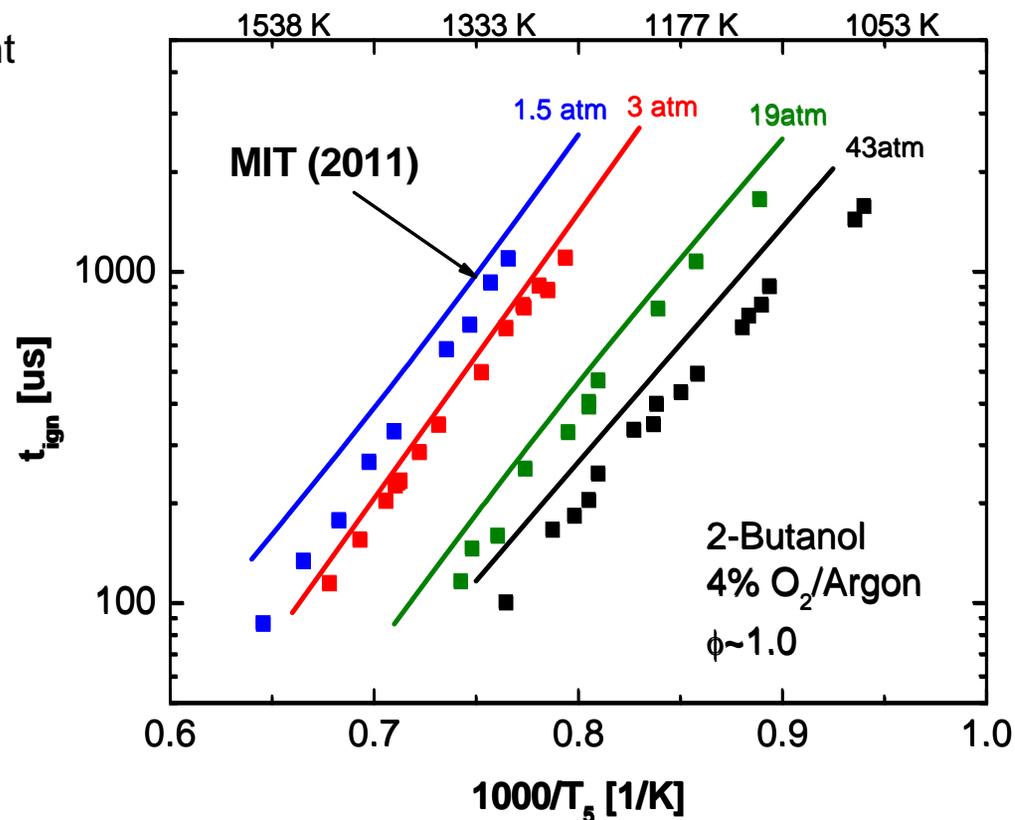
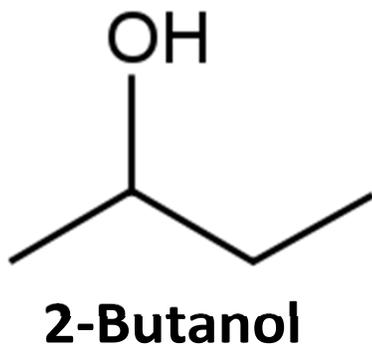
- **2-Butanol: 1-43 atm**
- **Methyl Oleate: 4-8 atm**

Ignition Delay Times: 2-Butanol Variation with Pressure in HPST

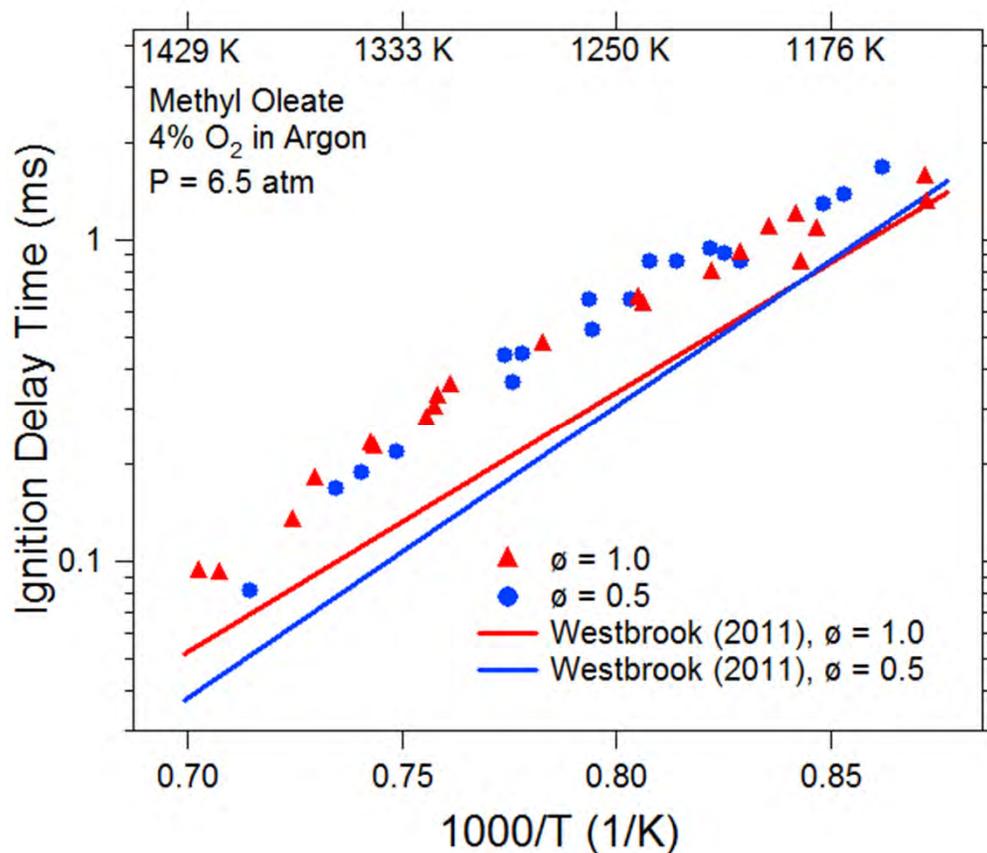
Very **low scatter** ($\pm 5-10\%$) consistent with uncertainty in T

Ignition delay times **scale approximately as $P^{-0.7}$**

MIT (2011) model P -dependence consistent with 2-butanol data
(but deviations larger than experimental uncertainty!)



Ignition Delay Times: Methyl Oleate in Aerosol Shock Tube

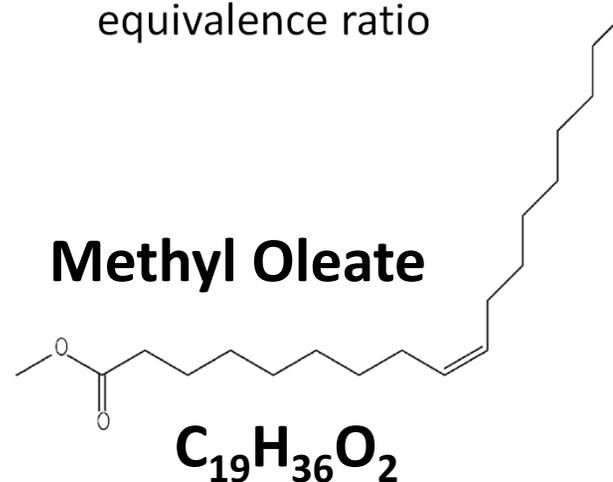


- Data reveal weak dependence on equivalence ratio

- **Westbrooke model:**

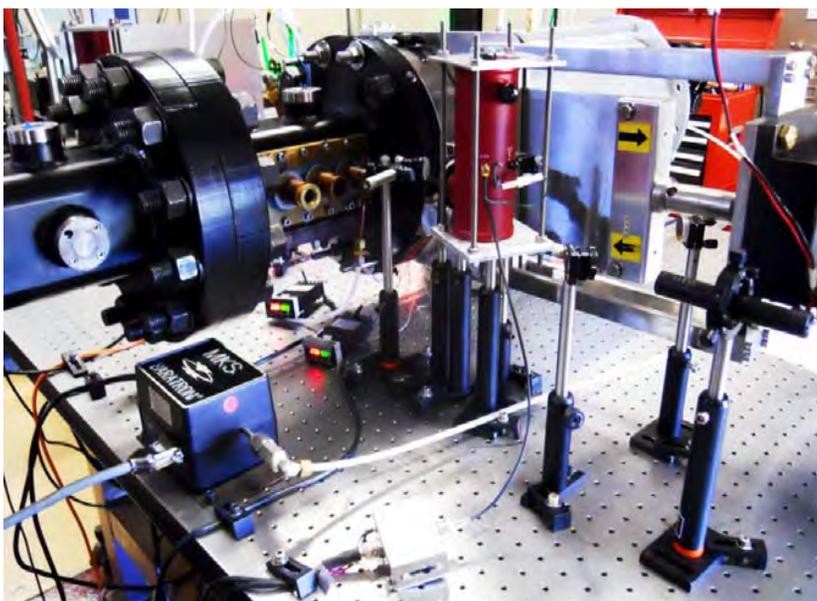
- underestimates ignition delay times by about 50%
- But confirms weak dependence on equivalence ratio

Methyl Oleate





Species Time-History Data at High P



**Aerosol
Shock Tube
10 atm**



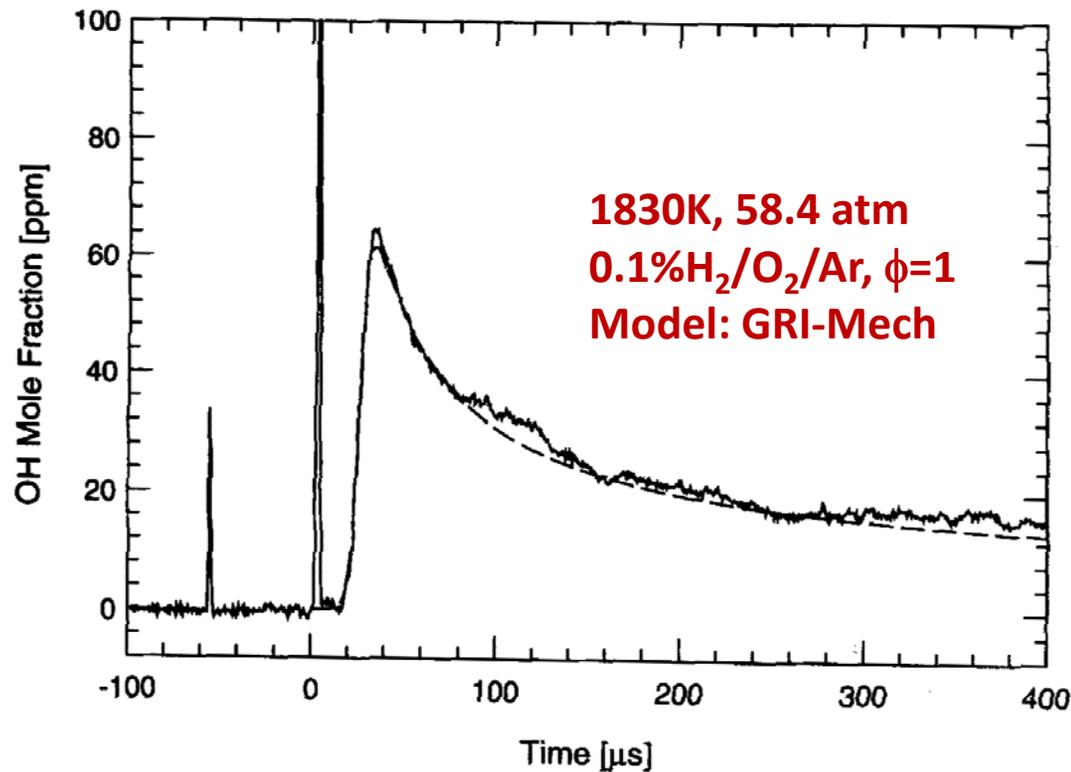
**High
Pressure
Shock Tube
500 atm**

Species Time-History Studies at Elevated Pressures

4 Representative Examples

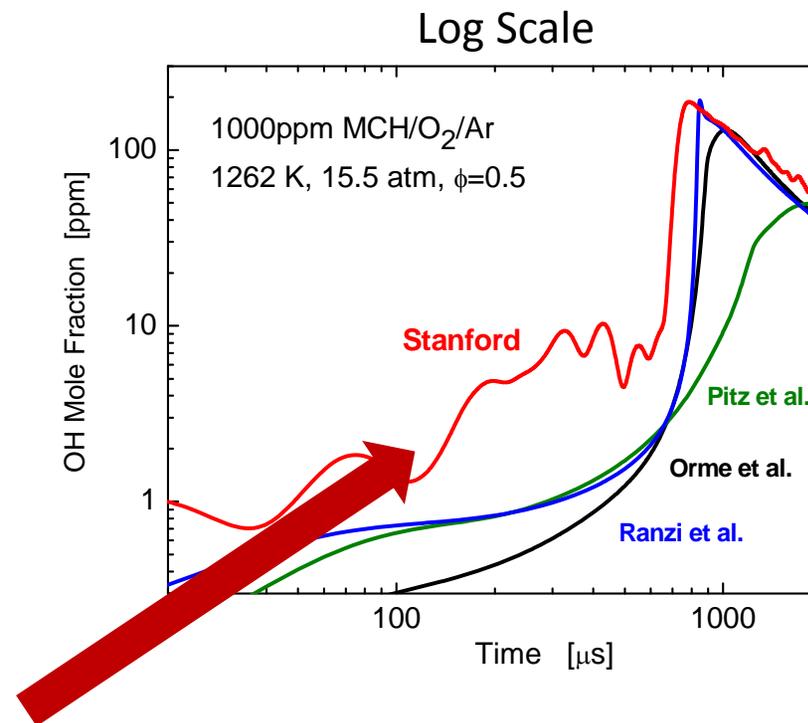
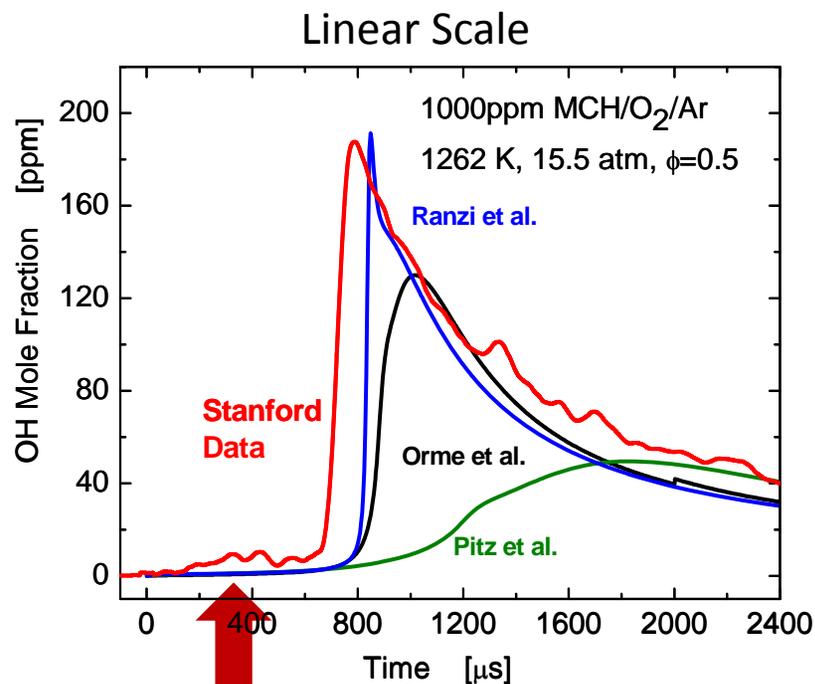
- OH species time-history during H_2/O_2 ignition (58 atm)
- OH species time-histories during jet fuel surrogate oxidation:
methylcyclohexane, n-dodecane, n-heptane (17 atm)
- C_2H_4 and Fuel time-histories during jet surrogate pyrolysis:
n-dodecane, methylcyclohexane, iso-cetane (15 atm)
- NO_2 emission time-histories during NO_2 pyrolysis:
Rate constant determination: $\text{NO}_2 \rightarrow \text{NO} + \text{O}$ (380 atm)

H₂/O₂ Ignition: OH Species Time-History



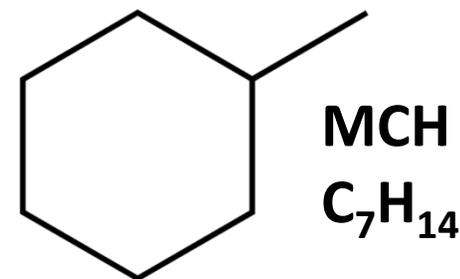
- High-quality, ppm-level detection of OH
- Useful for spectroscopic parameters/HP kinetics

Methylcyclohexane Oxidation: OH Time-History

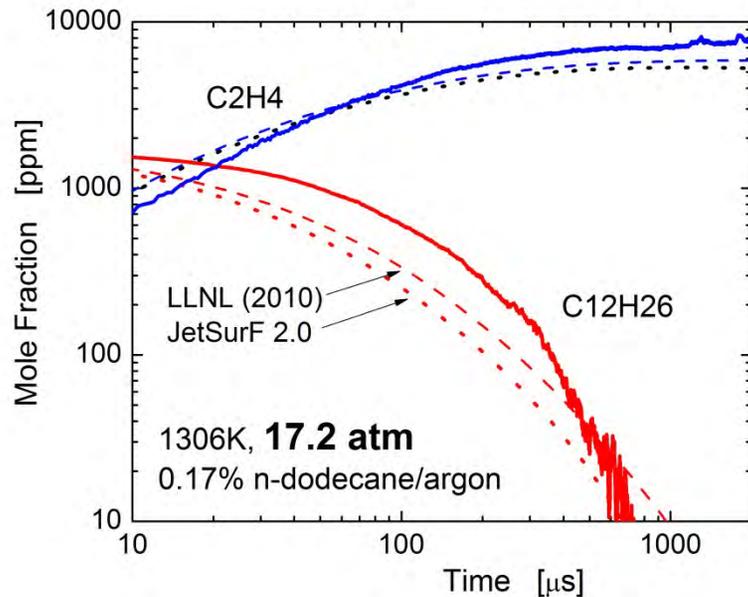


Early-time OH formation revealed

- Time-resolved speciation allows access to multiple phases of oxidation kinetics



n-Dodecane Pyrolysis: Ethylene and n-Dodecane Time-Histories

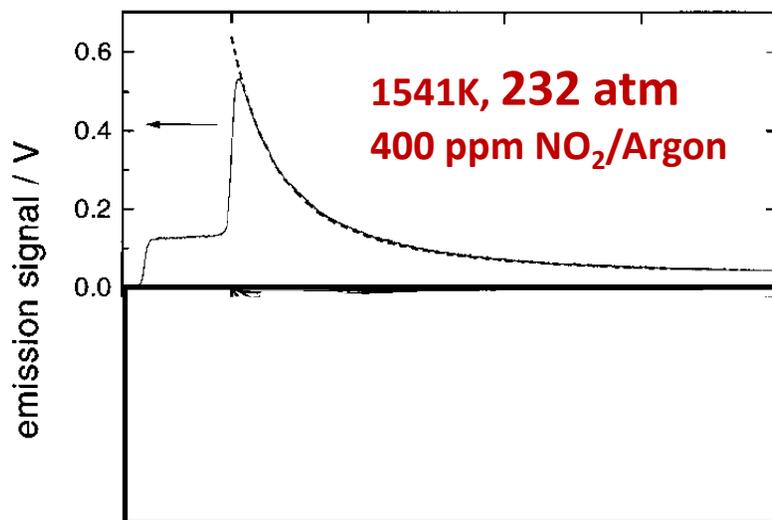


- IR laser absorption measurement
 - n-Dodecane (3.39 μm)
 - Ethylene (10.5 μm)
- C₂H₄ major contributor to carbon product balance
 - 78% conversion to C₂H₄
- Current n-alkane mechanisms do not capture time scales or ethylene yields at HP

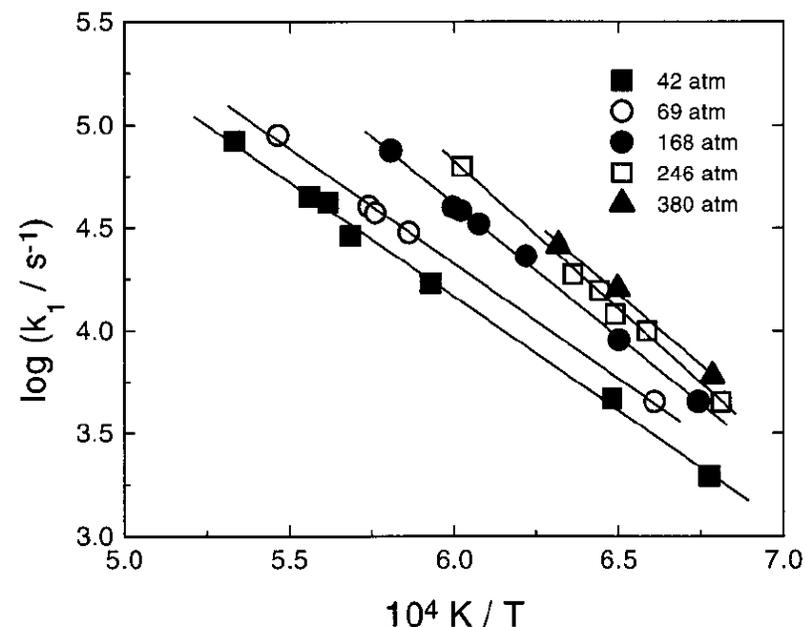
- **Species time-history measurements also provide access to direct measurement of elementary reaction rate constants!**

$\text{NO}_2 \rightarrow \text{NO} + \text{O}$ High-Pressure Rate Determination using IR Emission

IR Emission NO_2 Sensitivity



Arrhenius Plot



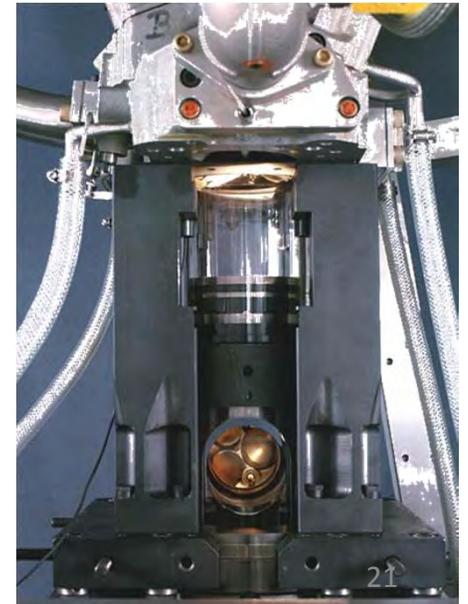
- Emission $\sim N_i^*$ \rightarrow improves with P!
- Real gas effects accounted for
- Sensitivity analysis shows NO_2 has strong sensitivity to title reaction

- Data obtained to 380 atm
- Then replotted for fall-off plot

Temperature Sensing

TDL sensors developed for variety of facilities and applications

- Shock tubes
- Coal gasifiers
- Engines



TDL Sensing at High Pressures

Issues/Challenges:

- Little spectrally-resolved absorption data at HP
- Absorption lineshape theory is inadequate at HP
- Diagnostic strategies needed for harsh environments
e.g. multi-phase extinction

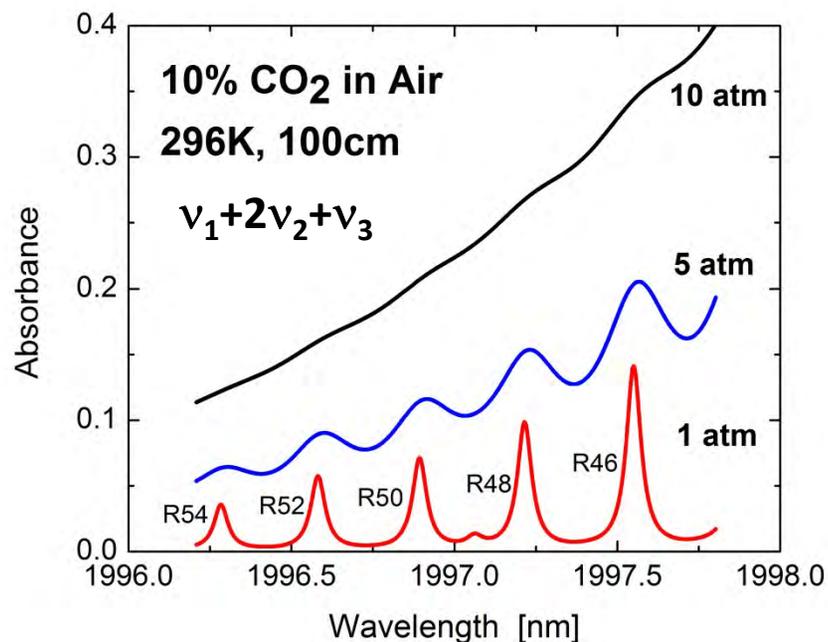
Strategies:

- 2f/1f WMS may provide a solution for moderate pressures (1-30 atm)
- Direct absorption of H₂O&CO₂ near 2.7μm for higher pressures (>30 atm)

3 Representative Examples:

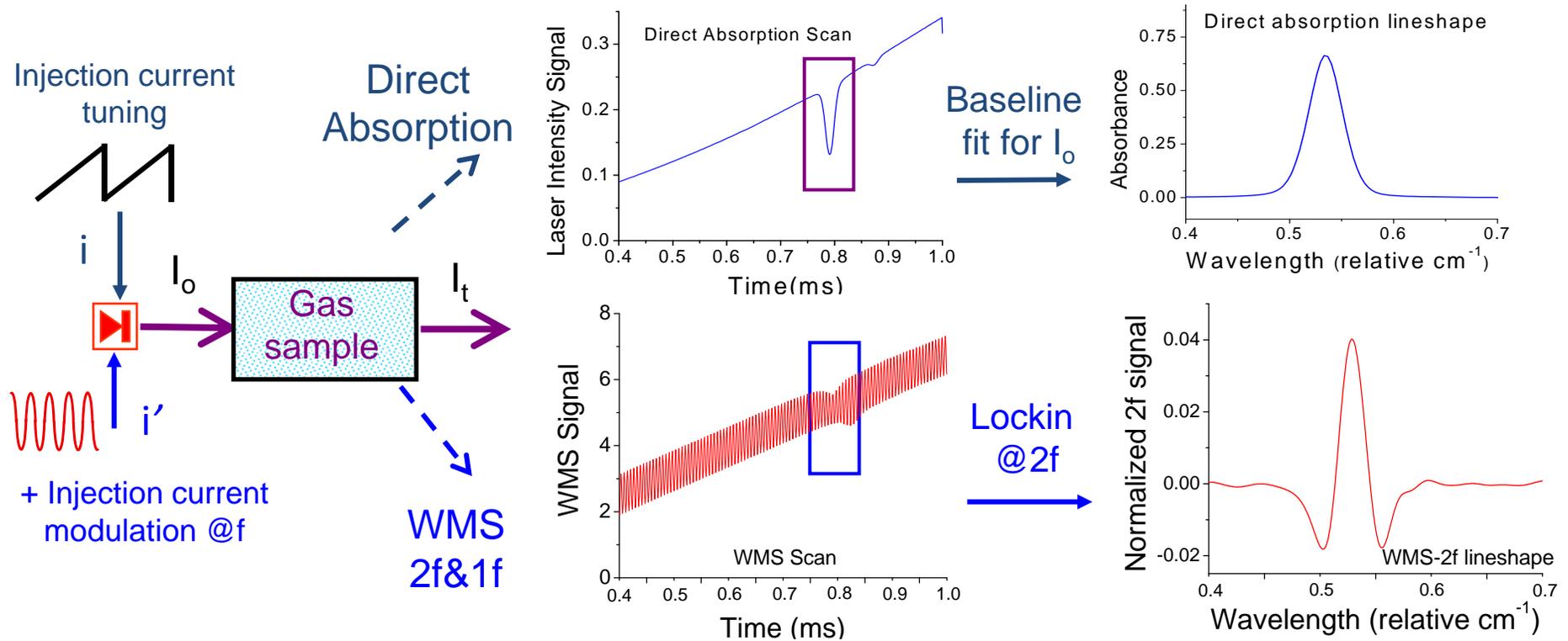
- **Temperature Sensing using CO₂**
- **Temperature Sensing using H₂O**
- **Simultaneous Species Time History and Temperature Sensing**

Problem: Spectral Character Changes with Pressure



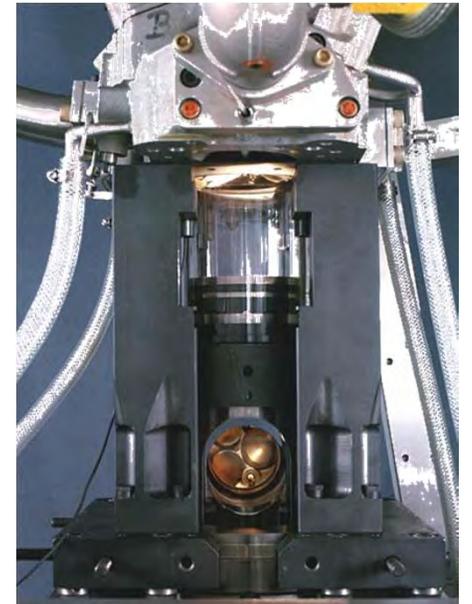
- Transitions broadened by collisions
- Transitions shifted by collisions
- Multiple absorption transitions blend; making species selectivity difficult
- Need for fundamental spectroscopy studies and “new diagnostic strategies”

Direct Absorption & Wavelength Modulation Spectroscopy

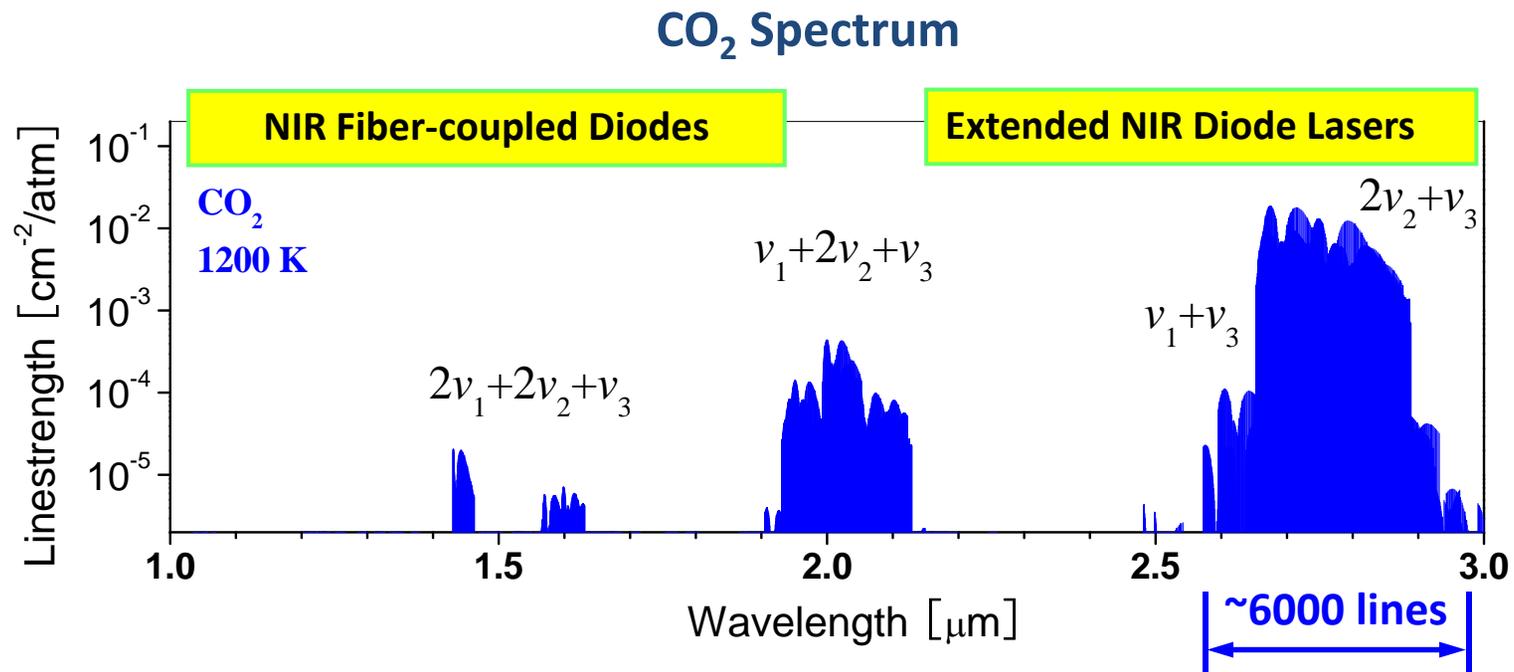


- **Direct Absorption:** Simple, if absorption is strong enough
- **WMS:** More sensitive, especially for small signals
 - Ratio of two WMS- $2f$ signals provides T (same as direct absorption)
 - WMS also produces intensity modulation @ $1f$
 - Both $2f$ and $1f$ signals proportional to intensity; $2f/1f$ **independent** of optical losses

High P Temperature Sensing Using CO₂



CO₂ Absorption



- 2.7 μm CO₂ bands are 30 \times and 1000 \times stronger than bands at 2 μm and 1.4 μm
- Stronger absorption: shorter sensor paths, lower detection limits, higher sensitivity

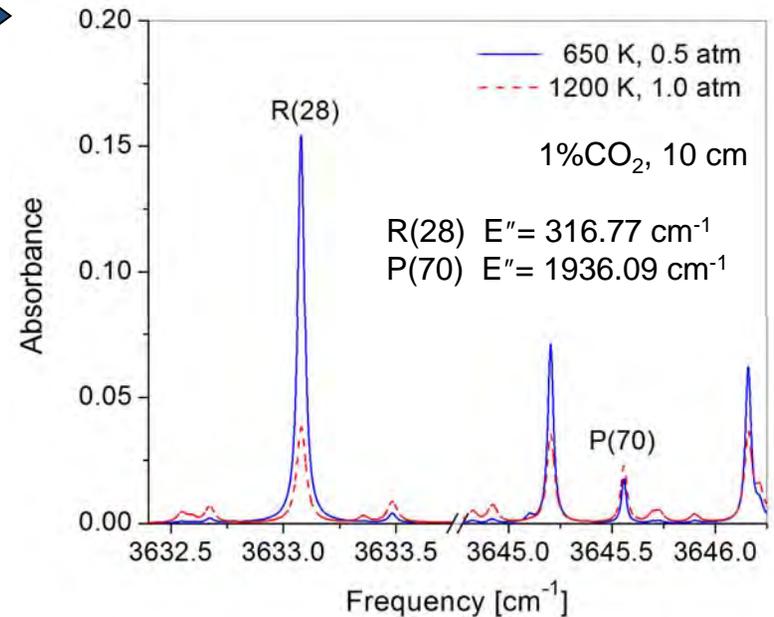
➔ For T, use ratio of two optimal absorption lines!

CO₂ T-Sensor Design

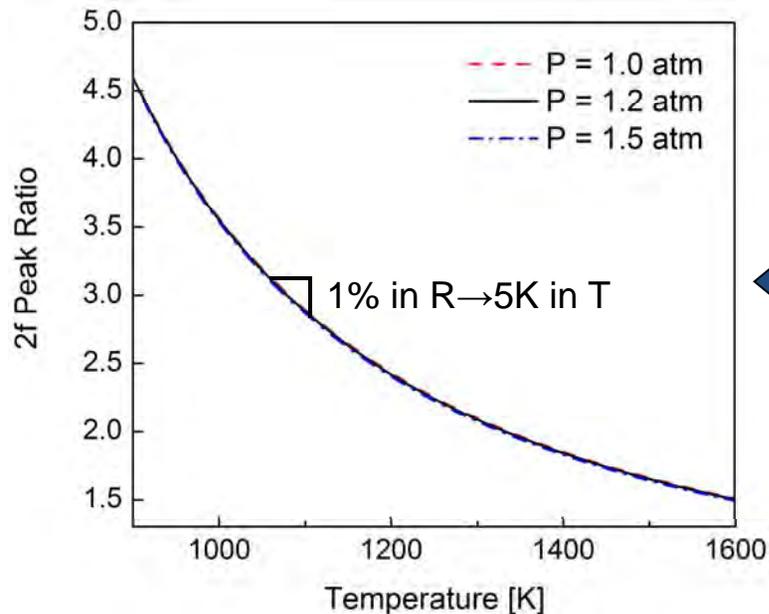
- **Optimal line-pair R(28) and P(70)**
 - Large line-strength $S(T)$
 - Isolated from neighboring lines
 - Well-separated lower state energy (E'')



CO₂ Line Selection

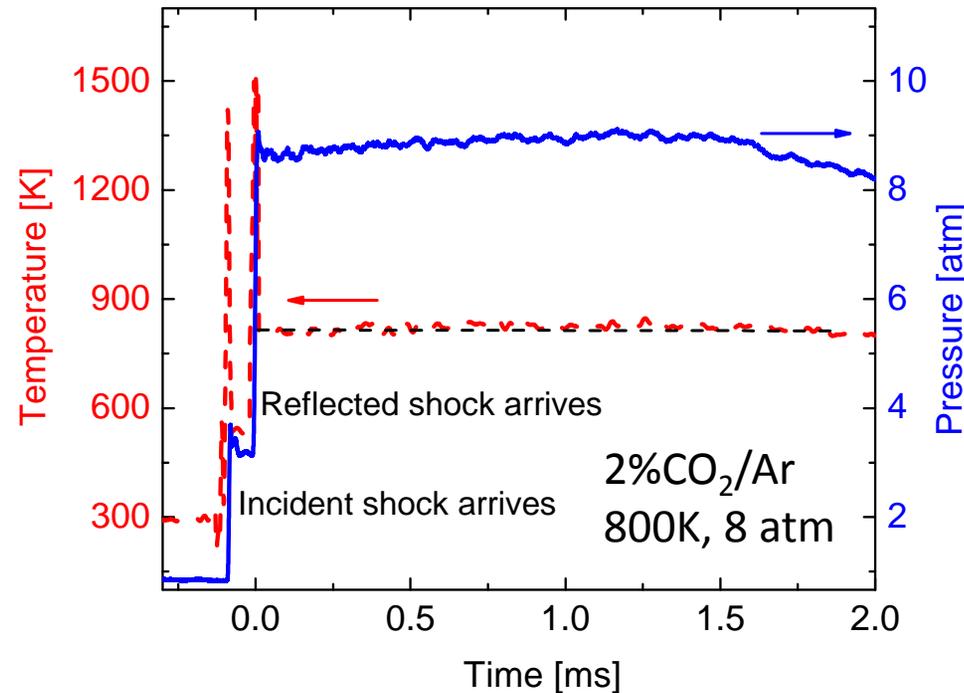


2f Ratio vs. Temperature



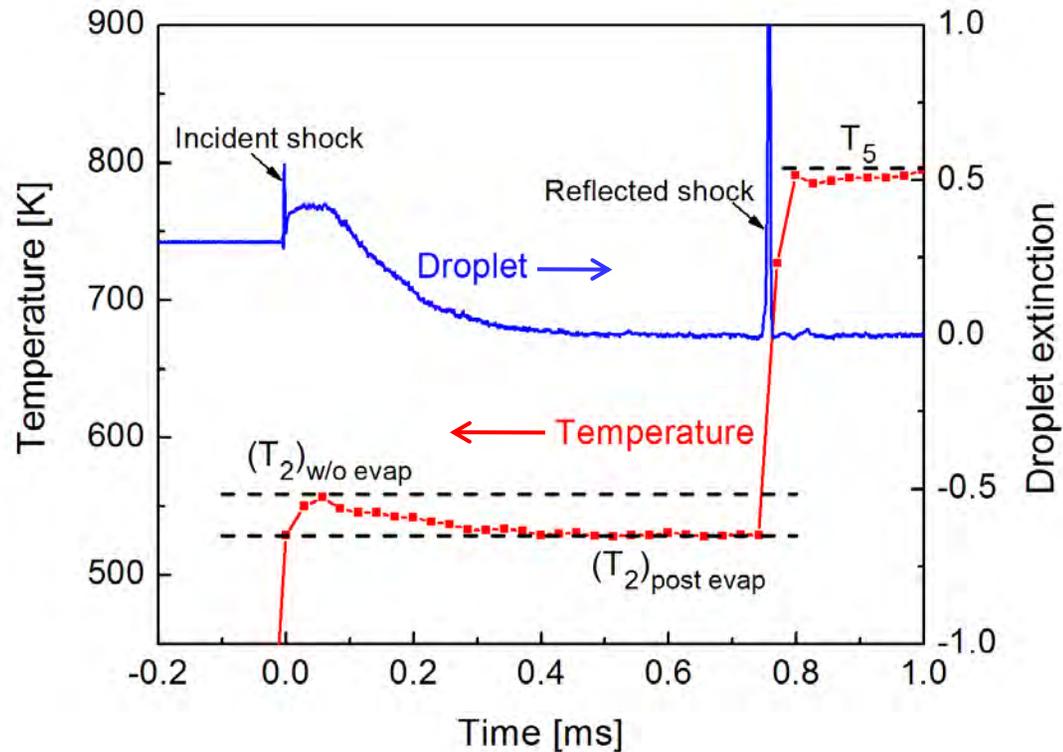
- **T inferred from measured 2f ratio**
- **Pressure influence is negligible**

Shock Tube CO₂ Temperature Measurement: 8 atm



- Sensor precision ~ 4K
- Shock tube attractive test bed for HP studies
- T measurement possible with CO₂ tracer or combustion products
- Successfully tested to 15 atm

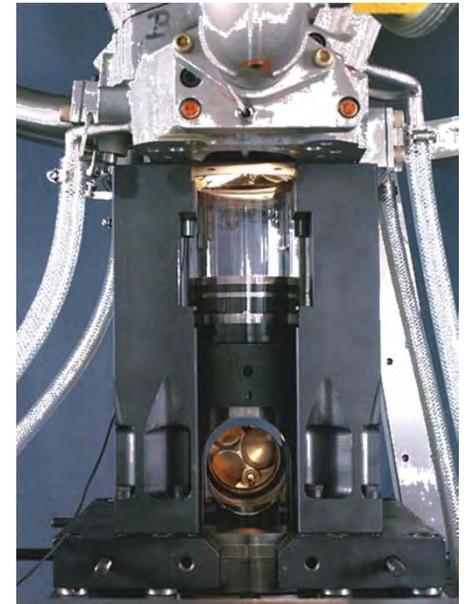
1f-Normalized WMS-2f for CO₂ with Particulate Scattering *Validated in Aerosol-Laden Gases*



**Aerosol shock tube:
2% CO₂/Ar, 0.5 atm
w/ dodecane aerosol**

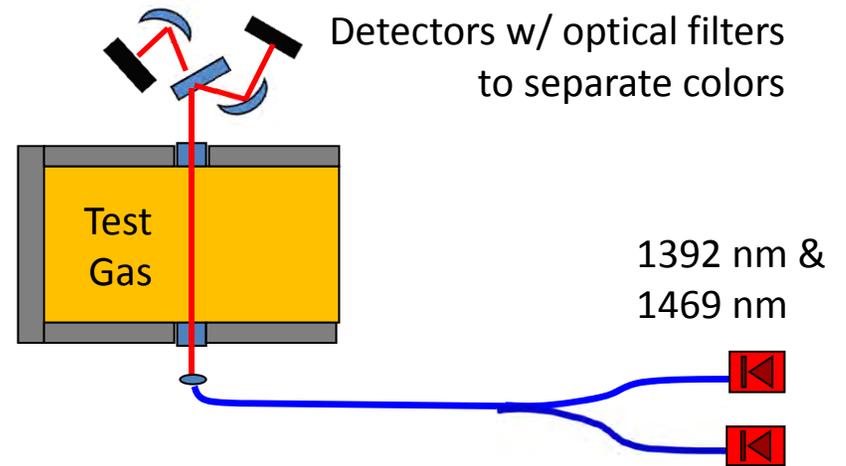
- **2f/1f TDL sensor successfully measures T in presence of aerosol!**

HP Temperature Sensing Using H₂O

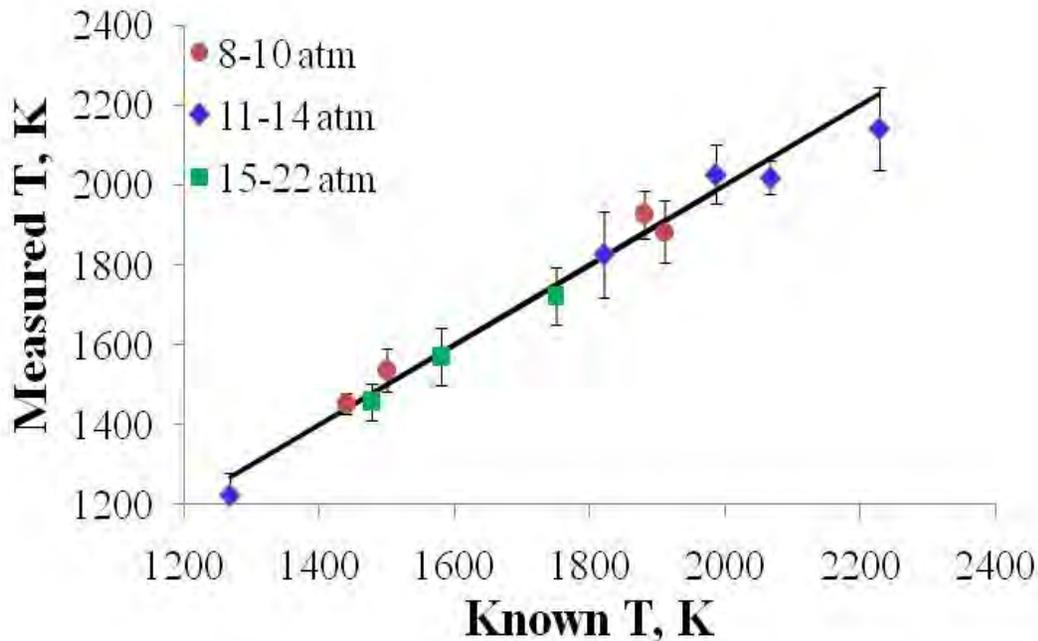


Temperature Diagnostic Validation in Shock Tube

- **Two-wavelength T sensor**
- **Direct measurement of T using H₂O in shock tube**



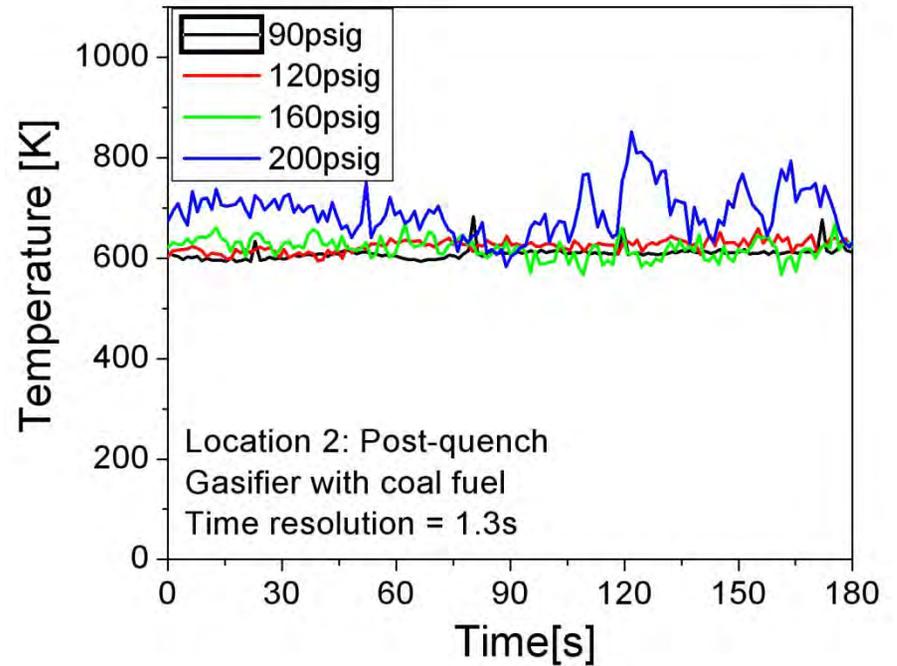
- **WMS Diagnostic validated to 22 atm and 2200 K**



Temperature in Gasifier: High P & Low Transmission

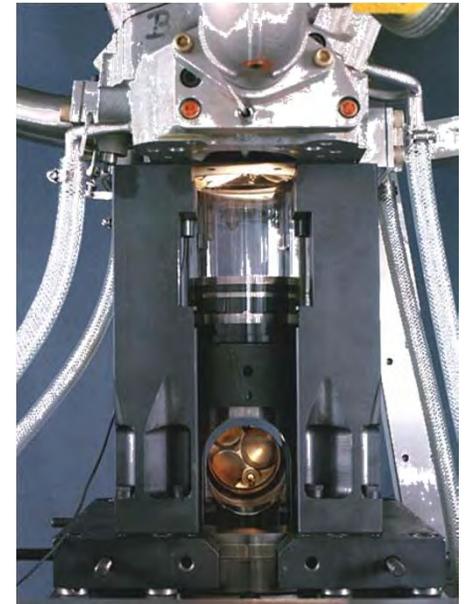


Sensor λ 's: 7426cm^{-1} , BW = 13kHz
 7466cm^{-1} , BW = 10kHz

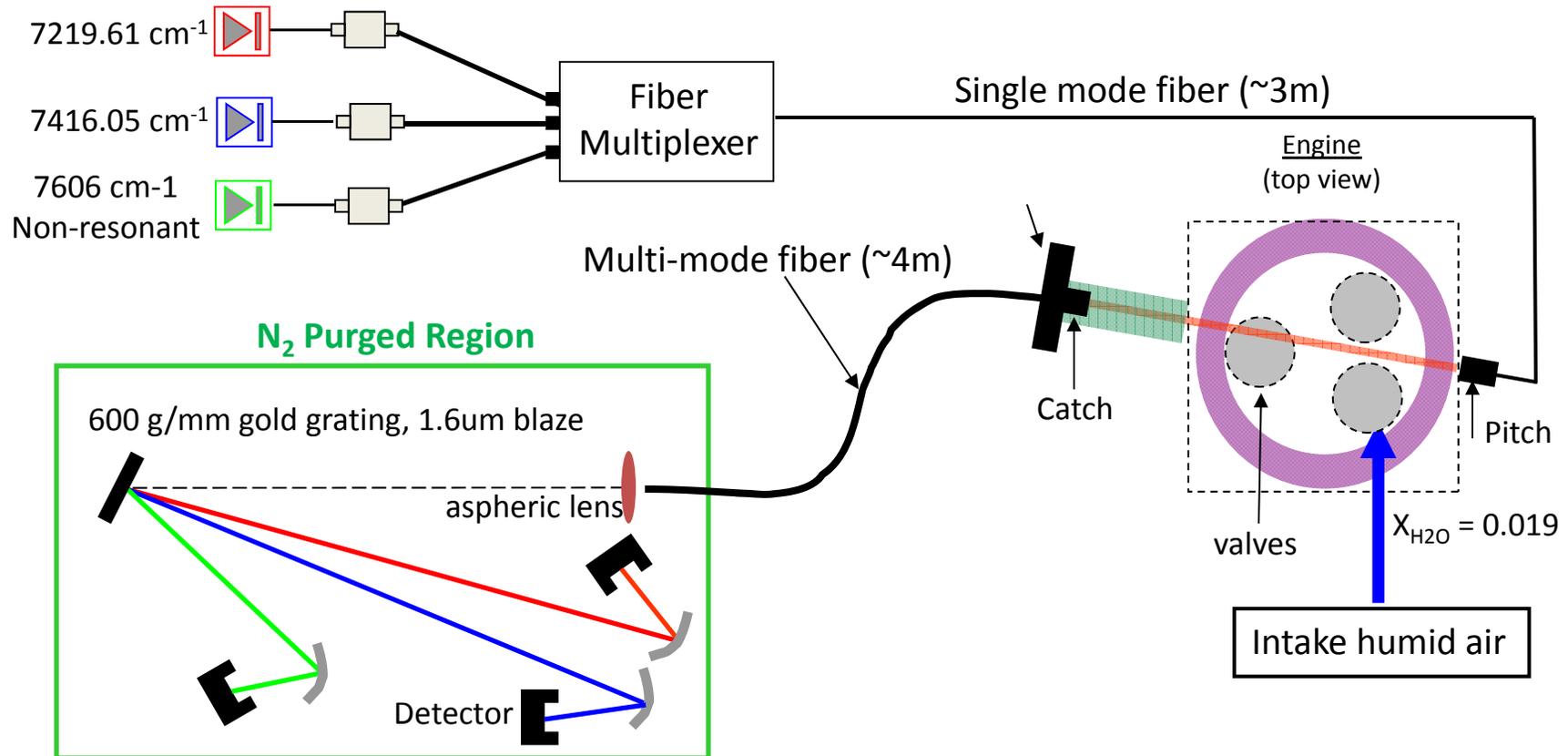


- High SNR, time-resolved measurements of T
- Normalized WMS enables T sensing even with transmission **less than 0.1%**
- Measured T at reactor pressures of 90, 120 and 160 psig stable
- Measured T at 200 psig identified instabilities (due to O₂ input flow)

Simultaneous Species Time-History and Temperature Sensing

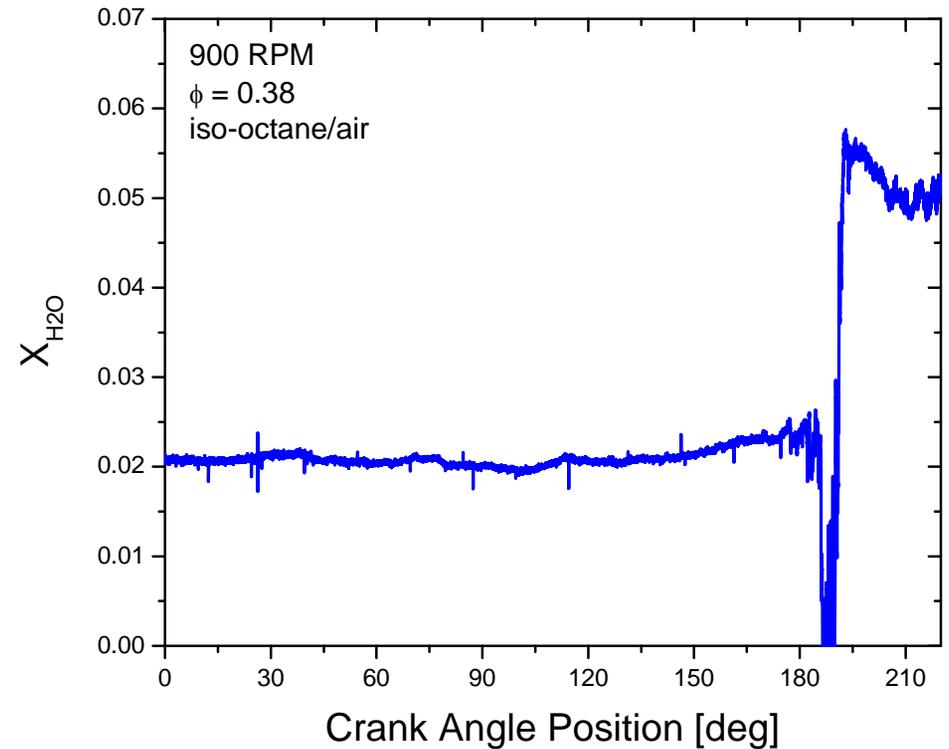
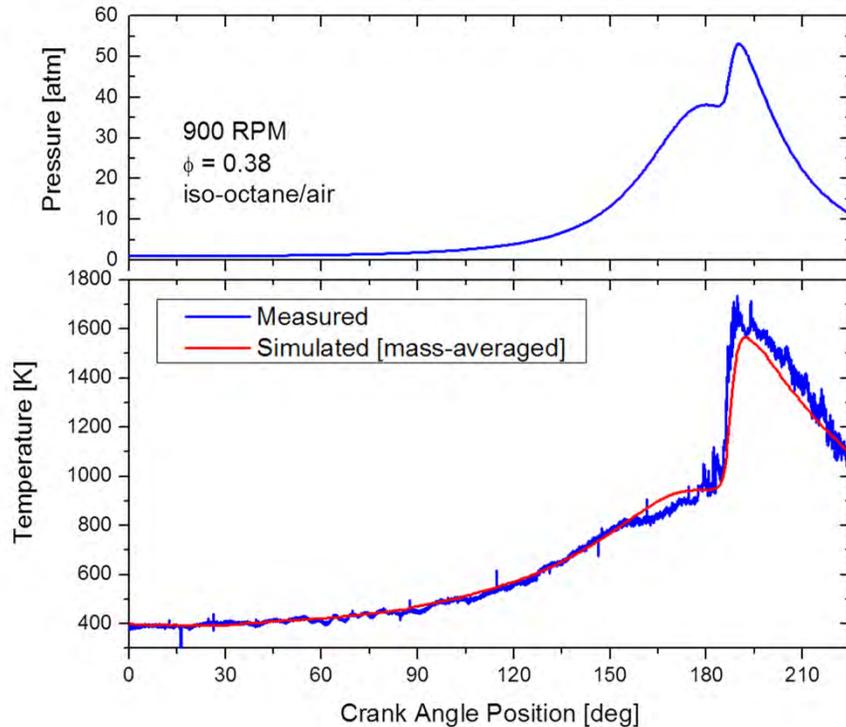


Cross-Cylinder TDL Sensor for Optical Engine



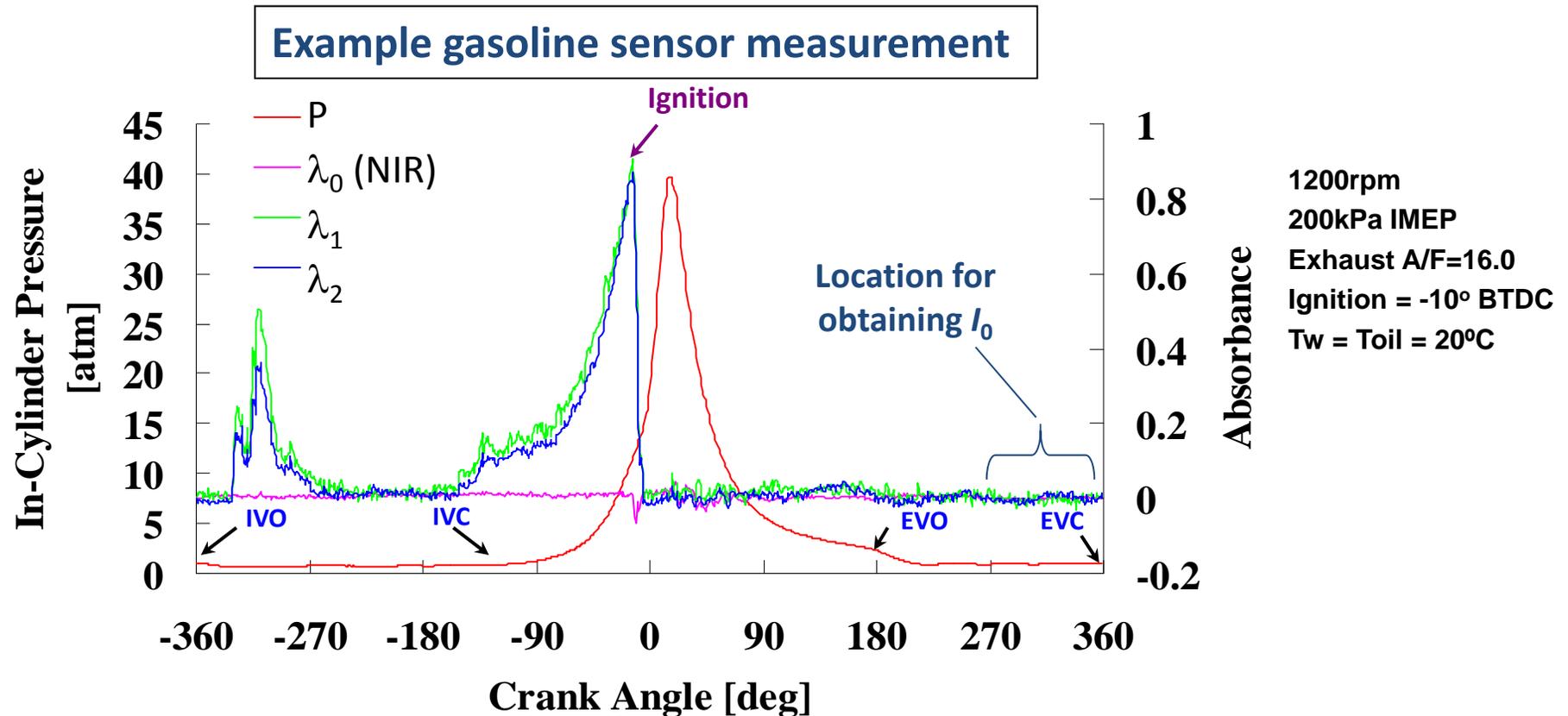
- Portable, robust, noise resistant, fiber-coupled optical system

Optical Engine Measurements at Sandia



- 2-color temperature; extra colors can be used to quantify non-uniformities
- Water mole fraction from T and absorbance
- Can be used to evaluate combustion models/determine combustion efficiency

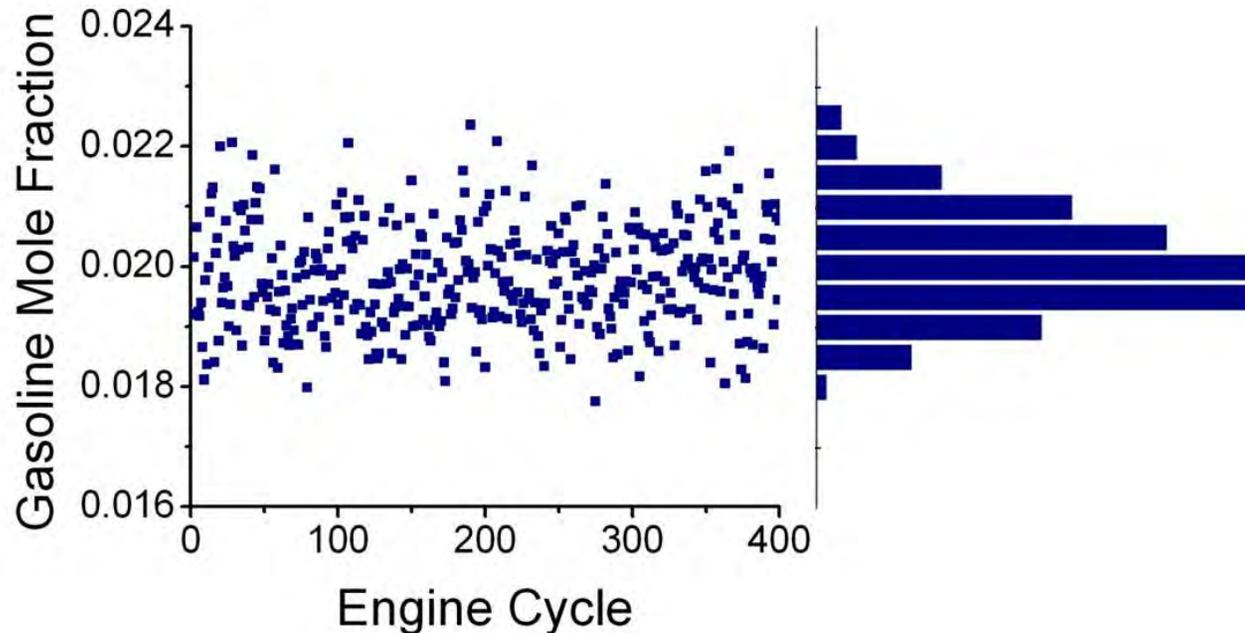
IC-Engine Application: Crank-Angle-Resolved Fuel/T Sensing with the Spark Plug Probe



- Single-cycle data from production engine
- High SNR with 5 mm probe gap!
- Sensor provides cycle-by-cycle results not available previously

Cold Start Ignition Statistics

Peak fuel at ignition for 400 cold start cycles



- High resolution quantifies small changes in fuel loading
- Sensor provides new tool to optimize MPI & DIG engines

Summary

- **Laser absorption well-suited to wide range of T & P**
 - Provides species time-histories at HP
- **WMS 2f/1f abs. allows measurement in harsh environments**
 - Even with 0.1% transmission or multi-phase flows
- **Need for quantitative studies of spectra at high P and T**
 - Current studies underway in UV, Vis & IR

Acknowledgements

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DOE, AFOSR, NSF, ARO, NISSAN